

Indoles in Multicomponent Processes (MCPs)

Morteza Shiri*

Department of [Ch](#page-36-0)emistry, Faculty of Science, Alzahra University, Vanak, Tehran 1993893973, Iran

CONTENTS

1. INTRODUCTION

Indoles are widely distributed in nature, many of which display important biological activities; moreover, a vast number of natural and synthetic indoles have found applications as pharmaceuticals and agricultural chemicals.¹ Therefore, the synthesis² and functionalization^{2b,3} of indoles have been the object of research for over one and half century[.](#page-37-0)

In the l[ast](#page-37-0) two or three decades the emphasis on and applications of green chemical principles, introduced some significant advances in organic synthesis, such as combinatorial chemistry, multicomponent processes (MCPs), organofluorine chemistry, organocatalysis, microwave synthesis and sonochemistry, etc. Among these new developments MCP techniques (including multicomponent reactions (MCRs) and one-pot multicomponent reactions) played a leading role and the field experienced tremendous developments.⁴ MCR is defined as a process in which three or more different starting materials react together and/or sequentially to yie[ld](#page-37-0) ideally a major product. Another variation of MCPs involves sequential addition of three or more different compounds in two or more steps to the same reaction vessel, entitled one-pot multicomponent reactions.

In the process consequent target molecules are isolated without purification and with minimal side products. Some advantages of MCPs involve minimization of the requisite reagents, solvents, cost, and time. Other advantages, which make MCPs an effective tool for synthetic chemists, are ease of separation and also minimizing the formation of chemical waste.

It appears that the term "MCR" was coined by Tsepalov in 1961.⁵ Nevertheless, the principle was already applied as far back as 1850 in a Strecker report.⁶ The first four-component reactions were developed in 1959 by Ugi et al. This appeared to have refreshed the interest in MCRs.⁷ Since then MCRs have found useful applications in the construction of valuable organic molecules as well as in the construc[tio](#page-37-0)n and functionalization of indoles.⁸ In this review, we wish to summarize the most important MCRs and one-pot MCR reactions in which either one of th[e s](#page-37-0)tarting materials or one of the product molecules are indole derivatives. The review will also present MCRs and one-pot reactions in the synthesis of indole alkaloids.⁹

2. FUNCTIONALIZATION OF INDOLES BY MC[P](#page-37-0)S

Rediscovery of the MCR one-pot reaction initiated a flood of investigations to determine the possibility to apply MCR technology to some of the traditional reactions that require two or three steps, leading to very exciting and valuable results. Carbon−carbon bond formation reactions in the field of alkylation of indoles were mainly targeted utilizing different pathways. A large variety of substituents were investigated.

Yonemitsu et al. was the first group to report the successful 3-CRs of indoles with Meldrum's acid 1^{10} and various aldehydes (Scheme 1).¹¹ Subsequent decarboxylative ethanolysis of adducts

Scheme 1. [Yo](#page-37-0)nemitsu Reaction

2 led to various ethyl 3-substituted indolyl-propionates 3 used as intermediates in the synthesis of complex indole alkaloids (Scheme 1).¹²

Liu et al. conducted this reaction by means of solid-phase synthesis and they us[ed](#page-37-0) a polymer-supported cyclic malonic acid ester, indole and aldehydes in the MCR.¹³ Upon completion of condensation reaction, the trimolecular adducts

Received: October 13, 2011 Published: March 5, 2012

in good yield and with high purity. As a result, the Yonemitsu reaction was successfully applied and extended to various complexes of alkyl- or phenylring-substituted indole derivatives, which proved to be a useful method in the synthesis of complex indole alkaloid precursors.¹⁴

For example, employment of Garner's aldehydes 4 in this reaction [with](#page-37-0) the corresponding indole derivatives 5 are converted through a series of different steps to the chiral 2′,3′-pyranone(pyrrolidinone)-fused tryptamines 6 and 7 (Scheme 2). 15

Scheme 2. [App](#page-37-0)lication of the Yonemitsu Reaction to Synthesize Chiral Tryptamines 6 and 7

Also Nemes et al. found that the Yonemitsu products 2 are hydrolyzed by t -BuOH to the hemiacid esters 8 (Scheme 3).¹⁶ The latter are transformed to carbamates 11 via a classical Curtius rearrangement mediated with diphenylphospho[ryl](#page-37-0) azide (DPPA). The intermediate corresponding acylazides 9, are subjected to a Curtius rearrangement to afford the isocyanates 10. Isocyanate intermediates 10 are converted to carbamate esters 11 by the addition of benzyl alcohol in high yields. Debenzylation of the carbamate esters 11 through catalytic hydrogenation produces the amino esters 12. Utilization of the Pictet-Spengler cyclization strategy with cyclohexanone affords cis- and trans-tetrahydro-carbolines 13 (Scheme 3).

Three-component condensation followed by cyclization of indoles, via the sugar aldehyde 14, and Meldrum's acid catalyzed by proline lead to $7-(1H-3-indolyl)-2,3$ dimethoxyperhydrofuro[3,2-b]pyran-5-ones 15 (Scheme 4).¹⁷ Expectedly, when an O-protected sugar is used, the noncyclized product 16 is isolated (Scheme 4). The reacti[on](#page-2-0) [of](#page-37-0) chiral sugar-derived aldehydes takes place in good yield and with high diastereoselectivity.¹⁸

Deb and Bhuyan^{14a} used barbituric acid 17 [in](#page-2-0)stead of Meldrum's acid in alternative strategy to [ob](#page-37-0)tain the Yonemitsu products 18 while bisindol[yl m](#page-37-0)ethanes (BIMs) 19 form as side products (Scheme 5). In the presence of protic solvent the yield of BIMs is increased.^{14a,19}

A plaus[ib](#page-2-0)le mechanism for the acid catalyzed-condensation of indole with al[dehyd](#page-37-0)es to BIMs 19 via the formation of

Scheme 3. Synthesis of Tetrahydro-carbolines

azafulvenium 20 are presented in Scheme 6^{3a} Azafulvenium 20 may be subject to additional nucleophilic attack by indoles or also to other nucleophiles pr[es](#page-2-0)[ent](#page-37-0) in the reaction media. It is noteworthy to mention that indoles and aldehydes in the presence of TFA/Et_3SiH give 3-alkylindole which is most probably converted to intermediate 20 via elimination of water.²⁰ However when indoles and aldehydes and imines are exposed to H_2 , 2,3- chiral disubstituted indolines²¹ and [2](#page-37-0)-alkyl indole 22 are isolated via an efficient Pd catalized tandem reaction. The reactions of indoles with ethers give [a](#page-37-0) variety of symmet[ric](#page-37-0) and unsymmetric 1,1-bis-indolylmethane derivatives via iron-catalyzed C−H bond oxidation and C−O bond cleavage.²³ In addition manganese chloride tetrahydrate was found to be effective for the selective transformation of indole, 3,4-dih[yd](#page-37-0)ropyran and $β$ -keto ester to the corresponding $BIM.²⁴$

The Yonemitsu methodology was extended to the synthe[s](#page-37-0)is of polyfunctionalized indole derivatives by a $TiCl₄/$ $Et₃N-promoted$ trimolecular condensation of aldehydes, indoles,and various activated carbonyl compounds by Renzetti et al. (Scheme 7).²⁵ Rationalization of these reactions and extension of it to other heterocyclic systems were also described.The same co[nd](#page-2-0)[ens](#page-37-0)ation of indoles, aldehydes and dimethyl malonate,²⁶ ethyl acetoacetate,²⁷ or nitroalkanes²⁸ were reported.

The variation in [p](#page-37-0)roduct distribution [of](#page-37-0) trimolecular co[n](#page-37-0)densation of indoles with aldehydes and methyl acetoacetate in the presence of $Ti(IV)$ and under different conditions were investigated.²⁹ Catalyzing the reaction with $TiCl₄/Et₃N$ or TiCl₂(O-iPr)₂/Et₃N gives only Yonemitsu product 21 but with $TiCl₄/Et₃N$ and extended reaction time, the tricyclic

Scheme 4. Sugar Aldehydes in Yonemitsu Reaction

Scheme 5. Formation of BIMs 19 as By-product in a Yonemitsu-type Reaction

 $cyclopenta[b]$ indole 22 is the dominant product (Scheme 8). It appears that the intermolecular Friedel−Crafts reaction of 21 is efficiently catalyzed with $TiCl₄$ to yield the tricy[cli](#page-3-0)c compound 22.

The reaction of β -dicarbonyl compounds such as 4-hydroxycoumarin or triacetic acid lactone, and indole in the presence of an aldehyde lead to the corresponding trimolecular condensed materials 23 and 24 respectively (Figure 1).³⁰ The reaction shows a surprising dependence on the solvent, with 1:1 chloroform-water giving the best yield of 23 and 24. [30](#page-3-0) [Ea](#page-37-0)rlier, the same method was employed to convert special 3-substituted indoles such as 23 and 24 to form the correspondi[ng](#page-37-0) adducts **25.** The product exhibits antibacterial activity (Figure 1).³¹

In a two-step sequential strategy involving green chemistry principles arylhydrazine reacts with methyl aroylacetate [to](#page-3-0) [fo](#page-37-0)rm 1,3-disubstituted 5-pyrazolones in glycerol at 110 °C for several hours. Treatment of indoles and paraformaldehyde under the

same conditions leads to the isolation of the corresponding indolyl derivatives $26-28$ in good yield (Figure 2).³²

Sapi and co-workers successfully prepared 1,2,3,4-tetrahydrocarbazoles 29 utilizing condensation of [2](#page-3-0)-[su](#page-37-0)bstituted indoles, Meldrum's acid and 2 equivalents of aldehydes in the presence of proline in high yields (Scheme 9).³³ This method was used for the synthesis of a large library of carbazols; their biological capacities were examined.³⁴

Perumal reported an efficient approach [in](#page-3-0) which a threecomponent condensation of salical[deh](#page-37-0)yde derivatives, malononitrile and indoles, catalyzed by $InCl₃³⁵$ or L-proline forms indolyl chromanes 30 (Scheme 10). 36 This protocol was extended to 2-hydroxynaphthalene-1-carb[oxa](#page-37-0)ldehyde, indole and malononitrile. The effectiveness [o](#page-4-0)f [th](#page-37-0)e reaction of indoles, aldehydes and malononitrile in aqueous media was reported by Qu et al.³⁷ Also the enantioselective version of reaction have been reported.³⁸

A metho[d](#page-37-0) for the synthesis of β -indolylketones 32 in good yields via con[de](#page-37-0)nsation of indoles, aromatic aldehydes, and

Scheme 8. Ti $(IV)/Et_3N$ -Promoted Trimolecular Condensation

Figure 1. Cyclic β-dicarbonyl in Yonemitsu reaction.

deoxybenzoin 31 with ultrasonic irradiation was described by Shen et al. (Scheme 11).³⁹ This one-pot process proceeds smooth and efficient in alkaline ethanolic solution.

3-(1-Arylsulfonylalkyl) [in](#page-4-0)[dol](#page-37-0)es 35 were generated from the three-component condensation of indoles, aldehydes or ketones, and arenesulfinic acids 33 (Scheme 12).⁴⁰ para-Toluenesulfonic acid-mediated azafulvenium 34 formation via condensation of carbonyls and indoles followed by t[he](#page-4-0) [add](#page-37-0)ition of arenesulfinic acids 33 sufficiently yield 67−95% 3-(1-arylsulfonylalkyl) indoles 35 (Scheme 12). The obtained products 35 undergo a Reformatsky reaction leading to alkyl 3-(3-indolyl) ketones 36.

The reaction of [iso](#page-4-0)xazole 37, aldehydes and indoles in the presence of BuLi leads to the formation of indole derivatives 40. The latter are easily hydrolyzed in aqueous alkaline media in a one-pot process to form valuable compounds such as indole 3-propionic acids 39 in good yield (Scheme 13).⁴¹ The

Scheme 9. Synthesis of Carbazoles 29

reaction most probably leads to the formation of styrylisoxazole 38 via activation of the methyl group in the oxazole 37 with BuLi followed by a condensation reaction with aldehyde. This method also works well for 7-azaindoles.

Alternatively, El Kaim et al. demonstrated that phenols react with aldehydes and N-benzylpiperazine 41 in toluene at 110 °C to give the Mannich adduct 42 as intermediate. The adduct can without isolation directly be converted to indolyl derivatives 43 in moderate yields when it is refluxed in toluene with 1,2 dibromoethane and 1-methylindole (Scheme 14). 42

In a sequential process hemiaminal ether 44 and enecarbamate 45 in the presence of the chira[l c](#page-4-0)a[tal](#page-37-0)yst (R) -46 (binaphthol derived phosphoric acid), form the hemiaminal product 46 which undergoes reaction with indole to afford the syn-product (48) in good yield and in nearly optically pure form; albeit with moderate syn diastereoselectivity (Scheme $15)$.⁴³ The extremely high enantioselectivity observed for syn-48 could be attributed to double asymmetric induction [aris](#page-4-0)i[ng](#page-37-0) from an ideal matching between the respective chiralities of the optically active product (47) and the catalyst (R) -46.

Condensation of indoles, formaldehyde, and tertiary aromatic amines catalyzed by silica-supported perchloric acid (HClO₄− $SiO₂$) to form dialkylaminoarylated indoles were described by

Figure 2. Structure of indolyl derivatives 26−28.

Scheme 10. Indolyl Chromanes 30 Formation Through **MCRs**

Scheme 11. Deoxybenzoin Formation in a Yonemitsu Reaction

Scheme 12. 3-(1-Arylsulfonylalkyl) Indoles

Scheme 13. One-Pot Synthesis of Indole 3-Propionic Acids 39 and Indole Derivatives 40

Kumar et al.⁴⁴ The proposed mechanism shows that the arylamine in the para-position is condensed with formaldehyde to give the i[min](#page-37-0)ium salt 49. The latter is then subjected to

Scheme 14. N-Methylindole and Mannich Products

nucleophilic attack by an indole unit that finally produces the 3-alkylated indole 50 in good yield (Scheme 16). The

Scheme 16. Mechanism for Indole, N,N-Dimethylaniline, and Formaldehyde Condensation

promotion of this reaction by ruthenium was reported,⁴⁵ and Liu et al. extended this reaction to other aldehydes using $FeCl₃$ as catalyst.⁴⁶

In a modification of the Mannich reaction (condensation of amines, aldehydes, a[nd](#page-37-0) activated aryls), Petasis successfully used aromatic or alkenylboronic acids instead of activated aryl compounds.⁴⁷ The Petasis reaction provides a selective approach to the synthesis of special molecules. After that indoles were pr[opo](#page-37-0)sed as substrates for the Petasis boronic acid-Mannich reaction, providing a practical synthetic route for

C−C bond formation in α -(N-substituted indole)carboxylic acids 51−56 from corresponding indoles, ethyl glyoxylic acid and aromatic or alkenylboronic acids (Figure 3).⁴⁸ In this

Figure 3. Products from condensation of indoles, aryl or alkenylboronic acids, and glyoxylic acid.

reaction a N-substituted indole replaces the amine component in the original Mannich reaction, yielding products in which two carbon−carbon bonds are formed during the multicomponent condensation.

Aryl-pyrrolo-tetrahydrocarbazoles 60 are isolated with ease from the three-component reaction of 2-substituted indoles 57, aldehydes and maleimides 59 in the presence of $CuSO₄$ as a catalyst (Scheme 17).⁴⁹ It appears that the diene 58, generated

from the condensation of indole 57 and aldehyde followed by H-shift,⁵⁰ reacts with the dienophile 59. By changing the starting materials and through some modification of the sequence [of](#page-38-0) the reaction a large diversity of compounds 60 was generated. These compounds disrupt protein−protein interactions and its activity was evaluated. 51

Similarly indole 61, 3,4,5-trimethoxyacetophenone 62 and maleimides 63 participate in a multicomponent reaction in protic media to yield tetrahydrocarbazoles 64 and the corresponding carbazoles 65 (Scheme 18).⁵² Tetrahydrocarbazoles 64 in the presence of dichlorodicyanoquinone (DDQ) ar[e](#page-38-0) oxidized to carbazoles 65 (Sc[hem](#page-6-0)e $18)$.⁵³ Ty et al. have investigated the biological activity of 64 and 65 in terms of antivascular action, the cytotoxicity against [mur](#page-6-0)i[ne](#page-38-0) B16 melanoma cells, the rounding up of endothelial cells (EA.hy 926) and the inhibition of tubulin polymerization.⁵³

The eco-friendly and one-pot Mannich reaction of aldehydes, primary amines, and indole[s i](#page-38-0)n water is catalyzed by $\dot{C_9}H_{19}COOH$ as a surfactant-type catalyst, 54 and was reported by Shirakawa and Kobayashi (Scheme 19).⁵⁵ Catalyst and solvent free version of this reaction but [un](#page-38-0)der thermal condition has been reported.⁵⁶

However, the majority of these reacti[ons](#page-6-0) involved the use of catalysts including ph[osp](#page-38-0)homolybdic acid (PMA) on silica gel under solvent-free conditions,⁵⁷ ZnCl₂ in ethanol,⁵⁸ acidic alumina under microwave irradiation,⁵⁹ acetic acid with reflux conditions, \rm^{60} ionic liquids, \rm^{61} Yb $\rm{(OTf)}_{3}$ $\rm{(OTf)}_{3}$ $\rm{(OTf)}_{3}$ -SiO $\rm{_{20}^{62}}$ bromo[dim](#page-38-0)ethylsulfonium bromide $(BDMS)_{0.5}^{63a}$ L-pro[lin](#page-38-0)e, $63b$ and also by 2,4,6-trichloro-1[,3](#page-38-0),5-triazine ([TC](#page-38-0)T) in $CH_3CN.^{64}$ $CH_3CN.^{64}$ $CH_3CN.^{64}$ Under high pressure conditions indoles [rea](#page-38-0)ct with [dich](#page-38-0)loromethane and secondary amines to give the correspondin[g](#page-38-0) Mannich products of 2-aminomethylindoles 66 in moderate to good yields (Figure 4).⁶⁵ Aldehydes in the presence of $LiClO₄$ react with $Me₃SiNMe₂$ to give the corresponding iminium salts that undergo [a](#page-6-0)[ddi](#page-38-0)tion of electron-rich aromatic compounds, leading to Mannich type products.⁶⁶ Zhang demonstrated the functionalization of resin-bound indoles utilizing the same process. 67

McFarland et al. used the Mannich product of 67, obtained from t[he](#page-38-0) condensation of 3-methylindole, formaldehyde and aniline in a phosphate buffer at $pH = 6.5$, as a model for characterization of proteins by isotopic labeling and NMR-based studies (Figure 5). 68 Adduct 67 is formed by the addition of 3-methylindole in position 1 to the imine from corresponding amine and for[mal](#page-6-0)[deh](#page-38-0)yde.

Similarly to this unique Mannich type reaction, the trimolecular condensation of 2-methylindole, benzaldehyde and ethanol in basic solution gives the ether 68 (Scheme 20).⁶⁹ The latter is unstable and is converted to the fulvalene 69 in freebase media (Scheme 20).^{3a}

3-CR of indole, ethyl glyoxylate 70 and anilines t[o](#page-6-0) [fo](#page-6-0)rm the acetate 71 was first r[epo](#page-6-0)r[ted](#page-37-0) by Passerini (Scheme 21).⁷⁰ Since valuable products are obtained through this method, it became quite a prominent reaction. It is well worth to [men](#page-6-0)[tio](#page-38-0)n the condensation of indoles, aniline and glyoxyl acid catalyzed with ytterbium triflate, 71 also under catalyst-free conditions in water 72 as well as solvent-free conditions.⁷

Secondary ind[oly](#page-38-0)l amines 72 are isolated in excellent yiel[ds](#page-38-0) from indole, 3-formyl chromon[e](#page-38-0) and corresponding anilines catalyzed by indium triflate under microwave irradiation while the corresponding bisindolyl methane 73 is obtained in poor yields (Figure 6).⁷⁴ Indium triflate was recovered after the reaction and reused in the subsequent experiments.

Zhang et al. [d](#page-6-0)[eve](#page-38-0)loped a highly enantioselective organocatalytic process employing the Friedel−Crafts aminoalkylation of indoles with imines generated in situ from trifluoroacetaldehyde methyl hemiacetal and aniline.⁷⁵ They employed the chiral phosphoric acid 74 as an efficient catalyst (Figure 7). A series of fluorinated indole derivativ[es](#page-38-0) 75−79 were obtained

Scheme 18. Synthesis of Tetrahydrocarbazoles 64 and Carbazoles 65

Scheme 19. Modified One-Pot Mannich Reaction

conditions: 8 kbar, 50 °C, 2-4 days

 $R^1 = R^2 = Et 68%$ $R^1 = R^2 = -(CH_2)_5 - 50\%$ $R^1 = R^2 = -(CH_2)_2 - 62\%$ R^1 = Me, R^2 = cyclohexyl 98%

Figure 4. Mannich products under high pressure conditions.

Figure 5. Structure of the Mannich product 67.

Scheme 20. Base-Catalyzed Condensation of 2-Methylindole, Benzaldehyde, and Ethanol

in high yields with excellent enantioselectivities (Figure 7). This methodology was further extended to include difluoroacetaldehyde methyl hemiacetal which enable a broad scope [o](#page-7-0)f new

Scheme 21. Indoles in a Mannich-Type 3-CR

Figure 6. Products of indole, 3-formyl chromone, and anilines.

substrates. This reaction also with another chiral binaphthol phosphoric acid 74 (9-phenanthryl instead of 2,4,6-(i- Pr ₃C₆H₂) also gives promising results.⁷⁶

Desimmoni and his group discovered that the reaction of aniline, indole and ethyl glyoxylate in t[he](#page-38-0) presence of $Sc(OTf)$ ₃ gives a mixture of complicated products such as 80, 81, and 82 (Table 1), depending on the reaction time and temperature.⁷⁷ As shown in Table 1, when aniline, indole and ethyl glyoxylate react at [−](#page-7-0)50 °C for 8 h mainly produced the product 80 in 9[5%](#page-38-0) yield (Table 1, en[tr](#page-7-0)y 1). The same reaction at room temperature and also after 8 h gives a lower overall yield, and 82 (48%) is the [ma](#page-7-0)jor product even though an appreciable amount of 80 (38%) is still isolated (Table 1, entry 2). The behavior of o-anisidine at different temperatures is quite different. After 30 min and at low temperature it [so](#page-7-0)lely gives the product 80, but at room temperature and after 3 h the ethyl 2-(4-amino-3-methoxyphenyl)-2-(1H-indol-3-yl)acetate $(82, R = OMe,$ Table 1) is isolated in 65% yield as the sole product (Table 1, entries 3 and 4). Successful rearrangements of 80 and 81 to 82 in the [p](#page-7-0)resence of $Sc(OTf)$ ₃ was also investigated.⁷⁷

This group also reported the MCR of indole, ethyl glyoxyla[te](#page-7-0) and 3,4-dimethoxy- or 3,4-methylenedioxyaniline[s w](#page-38-0)here the reactions are catalyzed by scandium triflate. This MCR gives two pairs of diastereomeric aza-Diels−Alder adducts (84 and 85). This appears to be a reaction in which the imines 83 behave as heterodienes and the indole as the dienophile (Scheme 22).⁷⁸

Figure 7. Asymmetric Mannich products via the chiral Brønsted acid 74.

The aza-Friedel−Crafts reaction of indoles with aldehydes and thiourea gives compound 86 in good to high yields. The reactions is efficiently catalyzed by thiamine/HCl or TiCl₄ (Scheme 23).⁷⁹ The thiourea derivative 86 forms various polysubstituted 2-amino-1,3-thiazoles 87 during the Hantzsch cyclizatio[n o](#page-8-0)f α -chloroketones (Scheme 23). The reaction is simple and allows for the introduction of at least three diverse functional groups which can be explore[d fo](#page-8-0)r the synthesis of multiple different compounds.

Ethyl glyoxylate, ethyl-4-amino-benzoate, and dihydropyridine-indole 88 unexpectedly yielded an indoloquinolizidine

Scheme 22. Povarov Adducts 84 and 85 from 3-CR

 $R^1 = R^2$ = OMe, overal yield 83% (ratio 84:85 (7:3)) in 0.5 h at rt $R^1 = R^2$ = OCH₂O, overal yield 75% (ratio 84:85 (85:15)) 1 h at -50 °C

derivative 89 isolated as a racemic mixture and in 66% yield (Scheme 24).⁸⁰

The tetrahydropyridinium ion 91 is generated from the $[4 + 2]$ cycloaddi[tion](#page-8-0) [of](#page-38-0) the N-alkenyl compound 90 and cyclohexene in a one-pot process. Further functionalization through nucleophilic addition of N-methylindole was achieved to yield the highly stereoselective adduct 92 in 53% (Scheme 25).⁸¹

2-Alkynylbenzaldehyde 93, p-toluidine and indole in the presence of $Ag(I)$, Cu(II) and Pd(II) effectively yield [1-\(](#page-8-0)1[H](#page-38-0)-indol-3-yl)-1,2-dihydroisoquinoline 94 but with $FeCl₃$ and triflate

Scheme 24. Intramolecular Approach to Indoloquinolizidine

Scheme 25. One-Pot Diels−Alder and Mannich Reactions

salts of $Zn(II)$, $Yb(III)$, $Bi(III)$ and $Dy(III)$ it solely gives the corresponding bisindolyl methane (BIMs) 95 (Scheme 26).^{82a,b} The reaction of 93 with different heteroarenes in the pres[ence](#page-38-0)

Scheme 28. 3-CR of 2-Alkynylbenzaldehyde 96, Malononitrile, and Indole

of bis(pyridine) iodonium tetrafluoroborate (IPy_2BF_4) gave functionalized 4-iodo-1H-isochromenes.^{82c}

As shown in Scheme 27, the synthesis of 1,2-dihydroisoquinoline 94 most possibly occurs in two ste[ps, f](#page-38-0)irst imine formation and then attack of the imine nitrogen to the alkyne followed by addition of indole to the carbon terminal of the imine group. $82a$ $Ag(I)$, Cu(II), and Pd(II) ions activate the alkyne in second step while several other salts are not effective catalysts.

In basic media, the indole nitrogen is deprotonated and then adds to the Knoevenagel adduct of aldehyde 96 and malononitrile (Scheme 28).⁸³ (Z)-1-Benzylidene-3-(1H-indol-1-yl)-1H-indene-2,2(3H)-dicarbonitriles 97 is selectively generated in moderate to goo[d](#page-38-0) yields via tandem condensation, nucleophilic addition, and 5-exocyclization.

Yadav discovered coupling of indoles to quinoline and isoquinolines which are activated by dimethyl acetylenedicarboxylate

Scheme 29. Coupling of Indoles and Isiquinolines

Figure 8. Compounds isolated from the condensation of dialkyl acetylenedicarboxylate, indoles, and isoquinoline or phenanthroline.

Scheme 30. 4-(Indolyl)dihydropyridines 107

at room temperature without a catalyst to produce indolyldihydroquinoline and isoquinolines (Scheme 29).⁸⁴ Isoquinolines, dimethyl acetylenedicarboxylate, and indoles give dimethyl (E)-2- [1-(1H-3-indolyl)-1,2-dihydro-2-isoquinolinyl]-[2-b](#page-38-0)utenedioates 98 in excellent yields and with high selectivity (Scheme 29). When a terminal acetylene and methyl propiolate are used the major product is an indolyl compound 99 (Scheme 29).

In contrast to this report, it was found that indoles are added to the 1-position of isoquinoline or phenanthroline with similar starting materials but through C−N bond formation.⁸⁵ For example structures 100−105 have been reported for the condensation of the corresponding dialkyl acetylenedicarb[ox](#page-38-0)ylate, indoles and isoquinoline or phenanthroline (Figure 8).

Pyridines in the presence of acetyl chloride and a proton sponge were activated to react with nucleophiles such as inScheme 31. N-Boc-azonino[5,4-b]indoles 109 via the Rearrangement of 108 with $(Boc)₂O$

Scheme 32. Mechanistic Rationale of the Decarbonylative Sonogashira Coupling

dole and is then converted to N-acyl dihydropyridines 106 (Scheme 30).⁸⁶ Subsequent treatment of 106 with alkaline methanol successfully yields (indolyl)dihydropyridines 107 in good to high [yie](#page-38-0)lds. Similarly 1,4-diazines react with indole in the presence of acyl halides. 87 Such reaction could be promoted in the presence of triflic anhydride.⁸⁸

Treatment of 108^{89} wi[th](#page-38-0) Boc anhydride as the acylating agent, in the presence of N-met[hyl](#page-38-0) indole as well as 2-(4 fluoro)phenyl indole[, a](#page-38-0)llows for access to the corresponding N-Boc protected azonino $[5,4-b]$ indoles 109 (Scheme 31).⁹ Formation of the intermediate 110 is plausible by the activation

Scheme 33. One-Pot Synthesis of α -Pyrrolidinyl Ketones

of 108 with $(Boc)_{2}O$; followed by addition of indole to 110, forming product 109 (Scheme 31).

Müller and co-workers demonstrated a new consecutive trimolecular preparation of alk[yno](#page-9-0)nes 114 via glyoxylation of indole, 7-aza-indole, and pyrrole derivatives with oxalyl chloride and subsequent Pd/Cu-catalyzed decarbonylative alkynylation of the heteroaryl glyoxylyl chlorides with terminal alkynes (Scheme 32). 91 The alkynones 114 are converted to pyrimidines in a subsequent transformation. The mechanistic path-way after [Fri](#page-9-0)[ede](#page-38-0)l–Crafts reaction of indole and $(COCl)_{2}$ and the oxidative addition of indole-3-glyoxylyl chloride to [Pd], adduct 111 involves a migratory deinsertion and elimination of carbon monoxide to furnish the acyl−Pd species 112 (Scheme 32). Then, transmetalation of the in situ generated copper acetylide to 112 gives rise to the formation of the acyl-alkyn[yl-](#page-9-0)Pd complex 113, which undergoes Sonogashira coupling (reductive elimination) to yield the alkynone 114. Previously in an alternative process starting from 3-iodoindoles, carbon monoxide and trimethylsilyl acetylene were also utilized for the preparion of similar compounds but in lower yields (see Figure 21). 92

Cernak and Lambert disclosed a new multicatalytic system in which both [an](#page-34-0) [am](#page-38-0)inochlorocarbonylation and Friedel−Crafts acylation reaction are utilized for the synthesis of α -pyrrolidinyl ketones (Scheme 33).⁹³ For instance aminoalkene 115 and indole 116 in the presence of $Pd(PhCN)_2Cl_2$, $CuCl_2$, $In(OTf)_3$ un[de](#page-38-0)r a carbon monoxide atmosphere give α -pyrrolidinyl ketone 117 in 72% yield (Scheme 33). Rh-catalyzed C−H carbonylation of indoles under 1 atm of CO has been achieved. 94 Various substituted indoles and indole with free N−H could be carboxylated with linear- or cyclic-alcohol to give the desir[ed](#page-38-0) indole-3-carboxylates with up to 92% yield.

In a Michael-type addition of nucleophiles to α , β -enals, an α -methylene group of the produced aldehyde 118 is activated for reaction with electrophiles leading to a series of new model compounds 119 (Scheme 34).⁹⁵

Scheme 34. Cascade Reactio[n o](#page-38-0)f α , β -Enals with Nucleophiles and Electrophiles

MacMillan's group used this basic reaction in an impressive design of an effective organocatalytic synthetic approach. They established the highly enantioselective synthesis of indole derivatives 122−124 by employing an indoles as the nucleophile and Cl⁺ as the electrophile (in situ generated from chlorinated

Figure 9. Products from a cascade of reactions via organocatalysis.

quinone 121). They used imidazolidinone 120 as the organocatalyst (Figure 9).⁹⁵ The Amine 120 mediated chiral catalytic cascade reaction via formation of an iminium/enamine is presented in Figure 1[.](#page-38-0)

After that, the group of Frechet proposed an iminium catalytic Michael [a](#page-3-0)ddition of N-methylindole to hexenal and enamine addition of the intermediate to methyl vinyl ketone as outlined in the Figure 10.⁹⁶ They found that the combination of immobilized catalysts 125 (the salt of imidazolidinonium with polystyrene sulfonat[e\)](#page-11-0) [and](#page-38-0) polymer supported diphenylprolinol methyl ether 126 sufficiently catalyzed MCRs of N-methylindole, hexenal and methyl vinyl ketone. Also by addition of a H-bonding catalyst such as 127 dramatically increases the yield from 33% to 89% while the enantiomeric excess stay >99%.

Similarly, another organocatalytic based reaction was reported by Melchiorre and co-workers where indoles, α , β enals and azocarboxylate react (Scheme 35). 97 Compounds 129 are efficiently prepared through the aminocatalytic activation of α , β -disubstituted enals. The use of cata[lyst](#page-11-0) [12](#page-38-0)8 in which chiral induction though a primary amine facilitates the fusion of these substrates into valuable precursors for α -amino acids with very high enantiomeric purity. The amino acids have two adjacent stereogenic centers.

In a related study, Arai and Yokoyama developed a threecomponent tandem reaction with indoles, nitroalkenes and aldehydes to construct acyclic products with three contiguous stereocenters 131 (Scheme 36). 98 They found that the asymmetric ligand 130 when complexed to CuOTf efficiently catalyzes the domino reacti[on](#page-11-0) [of](#page-38-0) a Michael-type addition of indole to nitrostyrene followed by the Henry reaction of the Michael-adduct with benzaldehyde. Diastereomer 131 is mainly isolated and 132 is also observed as minor product (Scheme 36). The addition of 1,1,1,3,3,3-hexafluoro-2-propanol

Figure 10. Polymer-based chiral catalysis for MCRs.

Scheme 36. Synthesis of a Three Contiguous Stereocenter Compound 131

(HFIP) was effective in enhancing the yield and to dramatically promote the stereoselectivity of the major product 131. They also demonstrated that reduction of the nitro group in 131, followed by Pictet−Spengler cyclization provided an efficient route for the synthesis of fully substituted chiral tetrahydro- β -carbolines (THBCs) 133 (Scheme 36).⁹⁹

Enders' group reported an innovative asymmetric organocatalytic scenario in which various indoles, nitroalke[ne](#page-38-0)s and 2 equivalents of acrolein are condensed to yield 3-(cyclohexenylmethyl)-indoles 134 bearing three stereogenic centers in moderate to excellent yields (23−82%) and with excellent stereoselectivities (dr = 91: 9 to >95: 5, ee = 94 to >99%) (Scheme 37).¹⁰⁰ The diphenylprolinol TMS-ether 135

Scheme 37. Retrosynthetic Analysis of the Domino Friedel− Crafts-type/Michael/Michael/Aldol Condensation Reaction

efficiently catalyzed this cascade reaction by an iminium/ enamine/iminium/enamine activation sequence.

A similar cascade reaction was reported by Kobayashi et al. Indole reacts with two different vinyl ketones in the presence of 5−8 mol % of Fe(BF₄)₂.6H₂O or Fe(ClO₄)₃.nH₂O to give 2,3dialkylated products 136-138 (Figure 11).¹⁰¹ The first alkylation step proceeds very quickly, while the reaction rate of the second alkylation is very slow.

Bandini et al. in an innovative strategy ha[ve](#page-12-0) [a](#page-12-0)dded indoles to the chalcones and then without purification, protected the Michael adduct with $Me₃SiCN$ (Figure 12).^{102a} This reaction is efficiently catalyzed by $InBr₃$ and opened a new route for the one-pot 1,4 and 1,2 addition of ind[oles](#page-12-0) [to](#page-38-0) α , β -unsaturated ketones. With the optimized atom-efficient protocol, several polyfunctionalized α -silyloxy cyanohydrins were synthesized in good to excellent yields (up to 97%) (Figure 12). Applicability of such reaction in the presence of nanocrystalline titanium(IV) oxide has been examined.102b

para-Toluenesulfonic acid (PTSA) efficie[ntly](#page-12-0) catalyzes the three-component couplin[g of](#page-38-0) indoles, styrenes and N-phenylselenophthalimide 139 yielding the corresponding seleno bearing indolyl compounds 140 in 63−95% yield (Scheme 38).¹⁰³

Also, the organophosphosphorus compound 141 is obtained in excellent yield from indole, dialkyl acetylenedicarb[oxy](#page-12-0)l[ate](#page-38-0) and triphenyl or trialkyl phosphite (Figure 13).¹⁰⁴ Although there are a few reactions of indole in the 1-position in this

Figure 11. 2,3-Dialkylated indole via cascade Michael addition.

Figure 12. One-pot synthesis of polyfunctionalized α -silyloxy cyanohydrins.

Figure 13. Examples of 1-indole type products.

 $context^{104}$ an appropriate example is the reaction of indole, dialkyl acetylenedicarboxylate and cyclohexyl isocyanide to produc[e th](#page-38-0)e aldimines 142 (Figure 13).¹⁰⁵

Alkynes activated by indium react with indoles and nucleophiles such as hydride, CN[−] generated from HSi[MeP](#page-38-0)h₂ and Me₃SiCN to form alkylindole 143 in 70–99% yield (Scheme 39).¹⁰⁶ 2-Methoxythiophene is also used as a nucleophile in this reaction. When the 3-position on the indoles is blocked, then positi[on-](#page-38-0)2 is functionalized. Also copper catalyzed reaction of indoles, sufonyl azides, and terminal alkynes led to the 3-functionalized indoles.¹⁰⁷

A three component tandem reaction for the synthesis of diversified N^a N^b dicarbamate-4,9-dihydro-3-iodo-a-carboli[nes](#page-38-0) 146 have been described by Sharma et al. (Scheme 40).¹⁰⁸ The reaction involves a one-pot condensation of bis-carbamate indole 144 with disubstituted propargyl alcohols 145, f[ollo](#page-38-0)wed by iodocyclo-elimination of the $N^{\rm b}$ -linked carbamate under mild conditions in the final step. In the presence of ICl the full aromatic form of 146 was generated.

Scheme 39. Indium-Catalyzed Alkylation of Indoles with Alkynes and Carbon Nucleophiles

The same group also reported several other iodinemediated MCRs yielding the indoloazepinone scaffold 3 via a three component reaction of indole-2-carboxamides 147, 1,3 disubstituted propargyl alcohols 148, and I_2 (Scheme 41).¹⁰⁹ The proposed mechanism for this reaction involves activation of the hydroxyl group toward electrophilic substitution r[eac](#page-13-0)t[ion](#page-38-0) by indole with iodine. Addition of iodine across the alkyne triple bond facilitates subsequent amide intramolecular cyclization to afford 149 in 55−72% yield (Scheme 41). Similarly, Rossi's group has reported the reaction of 2-acetyl indole, 1,3-disubstituted propargyl alcohols and amines in [th](#page-13-0)e presence of InCl₃ to yield 1-aminocarbazoles in $58-89\%$.¹¹⁰

Yavari and Habibi provided a novel procedure for the synthesis of 2-oxo-pyrrole and 3-oxo-indole derivatives [15](#page-38-0)4 (Scheme 42).¹¹¹ In this methodology isopropylidene Meldrum's acid 150 and isocyanide react via a $[4 + 1]$ cycloaddition forming t[he](#page-13-0) i[nter](#page-38-0)mediate iminolactone 151. The latter looses

Scheme 41. MCR Involving Indole-2-carboxamide, 1,3- Disubstituted Propargyl Alcohol, and I₂

Scheme 42. 3-Component Approach to Obtain 3-Oxo-indole Derivatives 154

acetone to give the acyl ketene 152. Addition of an indole to the lactone 152 induces cyclization of intermediate 153 to produce indolyl compounds 154 in 66−75% yield. Ren et al. have discovered that N-methylindoles are selectively transformed to their corresponding cyano derivatives in the 3-position via a palladium-catalyzed system in the presence of $NH₄HCO₃$ and DMSO.¹¹² It is worthy to note that the cyano group appears to originate from both DMSO and NH_4HCO_3 under these conditions. [Th](#page-38-0)is novel approach serves as a safe source for cyanide and it can potentially be used in other cyano addition reactions.

N-Substituted indoles in the presence of Selectfluor as the electrophilic fluorinating reagent and some nucleophiles such as H2O via difluorohydroxylation and alcohols via difluoroalkoxylation respectively led to the 3,3-difluoroindolin-2-ols 155− 157 and their alkoxy derivatives 158−160 in good yields (Figure 14). 113

3. SYNTH[ESIS](#page-38-0) OF INDOLE DERIVATIVES VIA MCPS

The MCRs technique caused significant advances in the construction of indoles, first by effective modification of traditional reactions such as the Fischer indole synthesis and second by planning new pathways for the synthesis of indoles. For example, a protocol described by Simoneau and Ganem involves formation of imines or ketones from nitriles or carboxylic acids. These intermediates then react with organometallic reagents and subsequent Fischer indole cyclization with an arylhydrazine hydrochloric acid salt to yield various

indole derivatives. 114 A modified approach to obtain N-arylindoles was developed by successive coupling of two different aryl bro[mid](#page-38-0)es to arylhydrazones, followed by cyclization with an enolizable ketone in a 3-component process.¹¹⁵

A novel tandem rhodium-mediated reaction involving hydrofo[rmy](#page-38-0)lation of olefins in an atmosphere containing both CO and $H₂$, followed by the Fischer indole synthesis with arylhydrazines 161, was the key to the synthesis of highly desirable indolyl structural motifs (Scheme 43).¹¹⁶

Scheme 43. One-Pot Hydroformylation and [Fisc](#page-38-0)her Indole Synthesis

Following this methodology it was found that if 1, 2-substituted alkenes were used, then indolenine intermediates 162 form. An acid-catalyzed Wagner−Meerwein-type rearrangement of the latter results in selective substitution at the 3-position. Selective substitution of the 2- and 3-position, lead to 2,3-disubstituted indoles 163 or 164 in good to excellent yields (Scheme 44).¹¹⁷

Another modification via titanium-catalyzed regioselective hydroamination [of a](#page-14-0)[lkyn](#page-39-0)es 165 with aryl hydrazines followed

by Fischer cyclization gives the indolyl hydrochloride salt 167 in good yield (Scheme 45).¹¹⁸ It is worthwhile to point out that

ammonia separates from hydrazine in the course of the cyclization. Substitution of chlorine is then subsequently achieved by ammonia floating in solution. Neutral substituted 3-(2-aminoethyl)- and 3-(3-aminopropyl)indoles 168 products are readily obtained when the salts 167 are washed in alkaline media. Three-component coupling between acyl chlorides, diazonium salts, and alcohols or amines allows the formation of α -hydrazono carboxylic acid derivatives which may be directly converted to indoles by means of a Fischer-type cylization.¹¹⁹ In addition, highly functionalized tryptamine derivatives via a Fischer indole type pathway using 2-methyl-1-pyrroline, ac[etyl](#page-39-0) chloride, phenylhydrazine, has been reported.¹²⁰

Müller and co-workers have established a novel method for the preparation of aryl substituted 5-(3-in[doly](#page-39-0)l)oxazoles 170 involving Sonogashira coupling of 169 and acyl chlorides, sequentially cycloisomerization followed by Fischer indole synthesis under microwave irradiation (Scheme 46).¹²¹ Interestingly aryl substituted 5-(3-indolyl)oxazoles 170 exhibit blue luminescence.

Scheme 46. One-Pot Three-component Synthesis of Indolooxazoles

Boruah and his group have developed a modified Nenitzescu reaction for the synthesis of 5-hydroxy-2,3-disubstituted-benzo- [g]indoles 171−174 (Figure 15).¹²² In this reaction naphthoquinone and acetophenones bearing aza-heterocycles are present, however the NH groups o[f ind](#page-39-0)ole originates from urea

Figure 15. Benzo[g]indoles formation.

under microwave irradiation without solvent. The reaction is efficiently catalyzed with BF_3OEt_2 . Very recently a multicomponent reaction leading to polyfunctionalized indoles and bis-indoles was established by Jiang et al. 123

A new MCR approach was revealed involving the one-pot reaction of gramine derivatives 176, o-alkenyl[phen](#page-39-0)yl isocyanide 175, diethylamine, and iodoarenes catalyzed by palladium (Scheme 47).¹²⁴

The palladium-catalyzed coupling of 2-alkynylisocyanobenzenes 178, allyl methyl carbonate and trimethylsilyl azide leads to N-cyanoindoles 182 (Scheme 48).¹²⁵ A plausible mechanism

involves the generation of π -allylpalladium azide 177 from allyl methyl carbonate and $TMSN₃$ (Scheme 48). Insertion of isocyanide 178 between the Pd−N3 bond in the π-allylpalladium azide 177 then yields the π-[ally](#page-14-0)lpalladium 179. Elimination of N₂ and 1,2-migration of the π -allylpalladium moiety from the carbon atom to the α -nitrogen atom lead to the formation of 180. The palladium-carbodiimide complex 181 forms via intermediate 180 and subsequent insertion of the alkyne moiety to the Pd−N bond. Finally at 100 °C the N-cyanoindoles 182 are isolated via intermediate 181 followed by the reductive elimination of Pd^0 . .

A novel multicomponent cascade process was reported by Gabriele et al. involving a sequential combination of an initial nucleophilic attack (ROH) to an imine moiety 183 and a palladium-catalyzed oxidative heterocyclization-alkoxycarbonylation process (Scheme 49).¹²⁶ In this new process, four simple

Scheme 49. Proposed Rea[ctio](#page-39-0)n Mechanism for the Formation of 1-(Alkoxyarylmethyl) indole-3-carboxylic

molecules [alcohol, carbon monoxide, alcohol, and oxygen] sequentially and selectively react with the 2-alkynylaniline imine starting structure, leading to high value-added functionalized indole derivatives 184 in a single operation catalyzed by oxygen activated Pd^0 . A chemo- and the regio-selectives tandem hydroformylation of substituted alpha nitrocinnamaldehydes leading to indoles has been reported.^{127a} Acetoxyindoles were obtained by cyclocarbonylation of 3-pyrrolyl acetates in the presence of $Ac₂O$ and a catalytic amount of [PdC](#page-39-0)l₂ under 50–70 atm of CO.^{127b}

In a careful study, construction and elimination of indole played an important role in the synthesis of highly fu[nctio](#page-39-0)nalized lactones (Scheme 50).¹²⁸ For example, the Ugi stereocontrolled condensation of acid 185, isocyanide 186, and p-methoxybenzylamine ($PMBNH₂$ $PMBNH₂$ $PMBNH₂$) leads to the formation of 187. Cyclization of the latter under thermal conditions forms the N-acylindole intermediate 188 and hydrolysis of the hindered amide 188 facilitates the preparation of the omuralide 189.¹²⁹ Rhoden et al. successfully used this strategy in the synthesis of N-substituted diketopiperazines.¹³⁰

A [fou](#page-39-0)r-component strategy was designed for the synthesis of polysubstituted pyrido[1,2-a]benzimidazol[e de](#page-39-0)rivatives 191 from 3-picoline 190, chloroacetonitrile, malononitrile, and benzaldehyde in refluxing acetonitrile. Interesting byproduct such as polysubstituted indole 192 forms in 8.6% yield (Scheme 51).¹³¹ The mechanism of this novel reaction appears to involve the

Scheme 50. Construction and Elimination of Indole in the Synthesis of Omuralide 189

Scheme 51. Polysubstituted Indole 190 as By-product in the Synthesis of the Benzimidazole 189

formation of polysubstituted benzenes with a subsequent substitution and annulation reaction of pyridine.

Condensation of acetylindol-3(2H)-one 193, malononitrile and aldehydes leads to 5-acetyl-2-amino-4-aryl-3-cyano-4H-pyrano- [3,2-b]indole 194 in the presence of Et_3N^{132} and NH₄OAc (Scheme 52).¹³³

Preparation of a wide variety of 3-aminoindoles 195 was accessed by Brønsted acid mediated imine formation of anilines and aldehydes followed by cyclization with tert-butyl isocyanide (Scheme 53).¹³⁴ When the aldehyde is changed to a ketone, the

Scheme 53. [3-A](#page-39-0)minoindoles

reaction leads to the preparation of substituted indoxyls. The similar reaction was previously reported by Deyrup et al.¹³⁵

Mossetti et al. during an attempt with an Ugi condensation found that 2-azidobenzaldehyde, phenylacetic acid, pen[tylis](#page-39-0)ocyanide and N-methyl benzylamine, led to the imide intermediate 196 in 75% yield. When the reaction is performed under aza-Wittig conditions, the 2,3-diamino indole derivative 197 is

Scheme 54. Formation of the 2,3-Diaminoindole Derivative

isolated in 70% yield (Scheme 54).¹³⁶ Aza-Wittig conditions with 2-azidobenzoic acid as starting material generate quinazolinones derivatives.

Alternatively 3-aminoindoles can be prepared using an efficient Cu-catalyzed three-component coupling reaction with 2-aminobenzaldehyde 198, a secondary amine, and an alkyne leading to a propargylamine intermediate 199. Under these reaction conditions 199 undergoes cyclization generating the indoline core 200 (Scheme $55)$ ¹³⁷ The latter, upon treatment

Scheme 55. Three-Component [C](#page-39-0)oupling Towards Indoles

with a base, smoothly isomerizes into the indole 201. The asymmetry version of this reaction with a chiral ligand complexed to copper was also reported in the same study.

A new strategy was reported in which a combination of a Ugi 4-CR and a Heck reaction was utilized. o-Bromoanilines, cynamaldehydes and isocyanides in the presence of formic acid form the intermediate 202. An intramolecular Heck reaction of 202 produces the desired indoles 203 in yields between 21 and 38% (Scheme 56).¹³⁸

A novel combination of the Smiles,¹³⁹ Ugi^{7c} and Heck¹⁴⁰ reactions was used f[or th](#page-39-0)e synthesis of indoles. In the Ugi-Smiles

Scheme 56. Combination of the Ugi and Heck Reactions

and Heck reaction ortho-iodonitrophenol reacts with aldehydes, amines and isocyanide to form 204. The latter in the presence of Pd(OAc)₂ gives indole 205 in good yield (Scheme 57).¹⁴¹

Scheme 57. Combination of Ugi-Smiles and Heck React[ions](#page-39-0)

Ackermann described the efficient construction of indoles 207 using the sequential coupling of ortho-dihaloarenes with phenyl acetylene and various amines (Scheme 58). The reaction proceeds by

Scheme 58. Employment of ortho-Dihaloarenes in Indole Formation

means of a multicatalytic system consisting of a N-heterocyclic car- $\mbox{binel}^{42\mbox{a,b}}$ or tri-tert-butylphosphine $\mbox{ }^{142\mbox{c}}$ palladium complex and CuI.

Barluenga discovered a new domino type three compo[nent](#page-39-0) strategy for the synth[esis](#page-39-0) of 2,3-disubstituted indoles (Scheme 59).¹⁴³ Primary amines in the presence of palladium and base are inserted into the bromoalkene 208 to give imines 209. The [latt](#page-17-0)[er in](#page-39-0) basic media generates the azaallylic anion 213 which couples with the dihaloarene 210 (Scheme 59). In the

Scheme 59. Pd-Catalyzed Construction of Indoles from Primary Amines, Bromoalkenes, and Dihalobenzenes

last step efficient intramolecular N-arylation of 211 occurs with the aid of the same catalyst to form the indoles 212 in 57−77% yield. High yield synthesis of furo[2,3-b]indole derivatives have been achieved utilizing 2-aminofurans formation via the reaction of isocyanide, cyclicketones and o-halobenzaldehydes followed by a copper-catalyzed intramolecular Ullmann reaction sequence.¹⁴⁴

The palladium-mediated reaction of 2-alkynyltrifluoroacetanilides with aryl halides in a carbon monoxide atmosphere le[ads](#page-39-0) to the isolation of 3-acylindoles 216. This reaction was first reported by Arcadi et al.¹⁴⁵ Similarly o -(o -aminophenylethynyl)trifluoroacetanilide 214 ($R = H$) reacts with aryl iodide 215 and carbon monoxide in [the](#page-39-0) presence of $Pd(PPh₃)₄$. To produce 3-acylindole 216^{146} (66%) along with the indoloquinazoline byproduct 217 $(17%)$ (Scheme 60).¹⁴⁷ Replacement of acyl

Scheme 60. Selective Synthesis of 3[-Ac](#page-39-0)ylindole versus 12-Acylindolo[1,2-c]quinazoline

amine with 214 (where $R = COCF_3$) leads to an improved yield of indole. 6-Trifluoromethyl-12-acylindolo[1,2-c]quinazoline 217 is formed in 89% yield (Scheme 60).¹⁴⁸ This method can also be applied to other aryl halides as well as triflates.

The methodology was also applied to the synthesis of pravadoline 218, an indole derivative designed as a nonacidic analogue of nonsteroidal antiinflammatory drugs (NSAIDs) (Scheme 61).¹⁴⁹ In the another study 2-aroylindoles using a

domino palladium-catalyzed C,N-coupling/carbonylation/C,Ccoupling sequence easily prepared from 2-gem-dibromovinylanilines and boronic acids under carbon monoxide.¹⁵⁰ If methanol be used instead of boronic acid the previous reaction the corresponding ester could be obtained.¹

Copper was found to be a suitable catalyst in the domino process for the formation of Manni[ch-](#page-39-0)products 221 from 2-ethynylanilines 219, formaldehyde and o-bromobenzylamines 220. Subsequent indole formation of 222^{152} follows, but if N-deprotection is introduced (to form N-arylation of 223) then the indole-fused benzo-1,4-diazepines [2](#page-39-0)24 are isolated (Scheme 62).¹⁵³ Nonstandard amines are used to construct

Scheme 62. [Cop](#page-39-0)per(I)-Catalyzed Domino Coupling− Cyclization−N-Arylation Reaction

more complex molecules such as carbolines,¹⁵⁴ tetrahydropyridine-fused indoles,¹⁵⁵ and benzoazepines¹⁵² with the same process. Au(III)-supported on $ZrO₂$ also effici[entl](#page-39-0)y catalyzes the reaction affording [com](#page-39-0)pound 222.¹⁵⁶ [An](#page-39-0) alternative method

Scheme 63. Synthesis of Aminomethylated Pyrroloindoles and Dipyrrolopyridines

that produces 2-(aminomethyl) indole derivatives 222 is the palladium−copper-catalyzed three-component assembling of propargyl halides, aryl halides, and secondary amines.¹⁵⁷

Ohta and co-workers have modified the route for the synthesis of pyrrole-fused indole derivatives 226, 227, [and](#page-39-0) 228 (when $X = CH$) utilizing a domino copper-catalyzed multicomponent coupling and bis-cyclization reaction (Scheme 63).¹⁵⁸ They discovered that the Mannich-type cyclization reaction of 4,6-diethynyl-1,3-phenylenediamine 225 ($X = CH$) can be c[on](#page-39-0)trolled with paraformaldehyde and a secondary amine to selectively form the mono- or bis-aminomethylated pyrroloindoles. The high-yielding bis-cyclization of terminal alkynes occurs in absence of Mannich-conditions to generate 227.

Chaplin and Flynn have described a one-pot, multicomponent coupling procedure for the synthesis of benzofuranes. They employed o-iodophenoles, alkynes and iodoarenes or alkenes.¹⁵⁹ They extended this methodology for the synthesis of 2,3 disubstituted indole 232. The reaction involves the Sonogas[hira](#page-39-0) coupling¹⁶⁰ of o -iodoacetanilide 229 to the terminal alkyne 230 followed by addition of a suitable coupling partner such as 231 to [give](#page-39-0) the indole 232 (Scheme 64). The same strategy

Scheme 64. One-Pot Synthesis of 2,3-Disubstituted Indole

was followed for the synthesis of a wide diversity of 2,3 disubstituted indoles by other groups.¹⁶¹ Very recently Rao et al. have established a method for the synthesis of indoles involving the one-pot coupling of (trim[ethy](#page-39-0)lsilyl)acetylene with iodoarenes in the presence of 10% Pd/C−CuI, followed by treatment of the reaction mixture with K_2CO_3 in aqueous MeOH, and finally coupling with o -iodoanilides.¹⁶²

SB 242784 (239), a compound in development for the treatment of osteoporosis, is synthesized from the coupling of alkynyl aniline 233 with bromoalkene 234 to form 235. A subsequent Suzuki coupling of 235 with 236 produces the ynediene 237 (Scheme $65)$.¹⁶³ Treatment of 237 with

 $Pd(CH_3CN)_2Cl_2$ induces cyclization to give the indole ester. Hydrolysis of the ester group and reaction of the resulting acid with amine 238 give SB 242784 (239).

Leogane and Lebel have developed the first sequencial onepot Curtius rearrangement/palladium-catalyzed indolization process for the direct synthesis of 2,3-disubstituted and 3-substituted indoles, as well as indole N-carboxamide derivatives 240, starting from readily available 2-iodobenzoic acid (Scheme 66).¹⁶⁴

A new and efficient copper(I)-catalyzed method was developed for the synthesis of 2-amino-3-alkylin[dol](#page-19-0)[e w](#page-39-0)ith the

Scheme 66. Synthesis of Indole N-Carboxamides

participation of 2-ethynylaniline 241, sulfonyl azide, and nitroolefin (Scheme 67).¹⁶⁵ As described in Scheme 67,

Scheme 67. Possible Mec[han](#page-39-0)ism for the Three-component Reaction of 2-Ethynylaniline, Sulfonyl Azide, and Nitroolefin

probably 2-ethynylaniline 241 reacts with the azide to form the reactive ketenimine 243 upon the ring-opening rearrangement of the triazole intermediate 242. Subsequently intramolecular nucleophilic addition occurs, leading to the intermediate 244. Afterward intermolecular Michael addition and tautomerization afford the desired product 246. HCT-116 inhibition activity of the produced indolyl compounds 246 was evaluated and it showed satisfactory results.

2-(2-Haloalkenyl)-aryl halides such as 247, react with amines with the aid of Pd catalysis to provide 1-substituted indoles.¹⁶⁶ When 1,3-dichloro substrate 247 reacts with different primary amines the corresponding three-component prod[uct](#page-39-0) 4-aminoindoles 249 and 250 in moderate yields were obtained (Scheme 68). A sequential metal-catalyzed C−N bond formation employing ortho-haloaryl acetylenic bromides and two different amines provides a facile access to 2-amido-indoles possessing a unique structural manifold.¹⁶⁷

4. PARTICIPATION OF INDOLE FU[NC](#page-39-0)TIONAL GROUPS IN MCPS

Using additional functional groups such as aldehydes, ketones, acids, amines, cyanides, etc., in MCRs makes it possible to prepare more complexes indole molecules. Mizoguchi et al. employed a Ugi 4-CR method to introduce indolyl derivatives 251. They used these compounds successfully as starting materials in the synthesis of poly fused cyclic compounds (Scheme 69).¹⁶⁸ Two patents used 3-formylindole and tryptamine in the synthesis of Ugi-adducts and evaluated the antibiotic activity o[f se](#page-39-0)veral bioisosteres.¹⁶⁹ N-Alkyloxazolidines react in a multicomponent reaction with carboxylic acids and isocyanides to give N-acyloxye[thy](#page-39-0)lamino acid amides.¹⁷⁰ Waki et al. have investigated the racemization in peptide synthesis using the Ugi-4CR of N-boc-3-formylindole, is[ocya](#page-39-0)nates and two differ-

Scheme 68. Sequential Amination Reactions

ent aminoacids.¹⁷¹ We found that ethyl 3-formylindole 2carboxylate participate in an Ugi-4CC with N-boc-amino acids to yield complex [dip](#page-39-0)eptides which are potential biological active molecules.¹⁷

Another strategy to increase the diversity of the indole scaffold is [to](#page-39-0) use bifunctional starting materials in which the participating functional groups of two components of the 4-CR are present in one structure.¹⁷³ With this idea in mind the bifunctional keto-acid indoles 252 (or pyrroles) react with isonitriles and amines to [form](#page-39-0) the corresponding novel 3-carboxamide derivatives 253 (Scheme 70).¹⁷⁴ The reaction

proceeds smoothly in methanol at 40 °C to yield the desired products in 64−92% yield. 3-Formyl-1H-indole-4-carboxylic acid is another example of a bifunctional molecule which undergoes Ugi condensation to give the corresponding heterocycles.⁸

Scheme 71. Indolobenzazepinones 255 and 257 via the Ugi Reaction

Beaumont et al. have demonstrated an intramolecular isocyanide-based multicomponent design utilizing the reaction of oxo-acids 254 and 256 to get access to indolobenzazepinones 255 and 257 respectively in good yields (Scheme 71).¹⁷⁵

A diversity of 1,4-thiazepine carboxamides 259, 261, and 263 were isolated utilizing a modified four-component [Ugi](#page-39-0) condensation reaction with bifunctional aldehyde/keto acids 258, 260, and 262, isonitriles, and amines (Scheme 72).¹⁷⁶

Scheme 72. Carbamoyl-Substituted Seteroannelated [1,4]Thiazepines

Considering the ease of the preparation of the initial starting materials, the convenient synthesis and isolation of the products, and the overall good chemical yields of the described transformations, this route provides a new and valuable entry to novel heterocycle-fused analogues of biologically active thiazepines.

The unique lactam structure 265 was isolated from 1,8 naphthaldehydic acid 264 upon reaction with tryptamine and phenylethyl isocyanide in 84% yield (Scheme 73).¹⁷⁷ A reaction between levulinic acid, isocyanides and primary amines in

Scheme 73. Indolyl Lactam 265

distilled water was reported in which γ-lactams derivatives were formed.¹⁷⁸ The reaction proceeds with the aid of a surfactant.

In an impressive design involving a six-functional group condensatio[n fr](#page-39-0)om a four component one-pot process, the tryptophanderived diketopiperazines 266 were obtained in good yield under heating or microwave conditions (Scheme 74).¹⁷⁹

Scheme 74. Synthesis of Diketopiperazines 266

The four-component nature of the Ugi reaction contributes to the formation of complex molecules with high diversity. The programmed combination of this multicomponent reaction with sequential secondary transformations has already been recognized as a powerful approach to get access to molecules with high molecular complexity. Indeed, many cycloadditions, cyclocondensations or organometallic couplings have been reported in which post Ugi condensations took place.

Ivachtchenko and his group established a Ugi-4CR for the construction of drug-like 2,3-dihydropyrazino[1,2-a]indole-1,4 diones 268 from indole-2-carboxylic acids, ethyl pyruvate, isocyanides, and primary amines (Scheme 75).¹⁸⁰ In this procedure

Scheme 75. 2,3-Dihydropyrazino[1,2-a]i[ndo](#page-39-0)le-1,4-diones 2 via a One-Pot Two-Step Procedure

after the Ugi condensation of four mentioned compounds to form compound 267 at room temperature at 50 °C in methanol, the solvent is removed in vacuo and the resulting solid is redissolved in glacial acetic acid. The cyclized product 268 is isolated in moderate to good yield when the sealed reaction vessel is heated under microwave irradiation.

Construction of 269 via N-arylation of 270 would be access (Scheme 76). Diamide 2 is Ugi adduct indole-2-carboxylic acid,

Scheme 76. Combination of Ullmann and Ugi Reactions in the Construction of Indolo $[1,2-a]$ quinoxalinones 269

aldehyde, isocyanide and 2-iodoaniline (Scheme 76). Balalaie and co-workers by this strategy gave a practical method which came from combination of Ullmann N-arylation and Ugi 4CC to preparing indolo $[1,2-a]$ quinoxalinones 269 in good overall yield (Scheme 76).¹⁸¹ Ugi reaction was performed in methanol and N-arylation done in the presence of CuI/L-proline.

Dömling and hi[s gr](#page-39-0)oup used an Ugi-deprotection-cyclization strategy for the preparation of 1,4-thienodiazepine-2,5-diones 272 from 2-amino-4,5,6,7-tetrahydrobenzo[b]thiophene-3-carboxylic acid 271, an α -amino acid, isocyanate and formaldehyde (Scheme 77).¹⁸² For example when tryptophan (as the α -amino

acid) and t-butyl isocyatate is used in this reaction the indolyl derivative 273 is isolated in 16% overall yield. The mechanism involves an Ugi condensation in methanol, followed by amine Boc-deprotection in TFA and treatment with triethyl amine in third step (Scheme 77). These hybrid peptidomimetic diazepine structures were evaluated for chemical space distribution and druglike properties.

The Ugi−Hulme-modified condensation of monoboc protected ortho-phenylene diamine, glyoxylic acids, aldehydes, and isocyanide in a two-step procedure has produced several quinoxalinone derivatives (Figure 16).¹⁸³ In the first step Ugicondensation occurs followed by simultaneous boc-deprotection

Figure 16. Indolyl quinoxalinones.

and cyclization in the presence of TFA in dichloroethane to yield the target molecules.

A two-step procedure involving an Ugi reaction preceding a Pictet−Spengler reaction allows for the isolation of the tetrahydro- β -carboline scaffold 274 (Scheme 78).¹⁸⁴

Scheme 78. Combination of the Ugi and Pictet[−](#page-40-0)Spengler **Reactions**

Alternatively, employment of bifunctional materials that have both aldehyde and acid groups react with aminoacetaldehyde dimethyl acetal and 2-isocyanoethylindole to afford some interesting polycyclic compounds such as 275−278 (Figure 17).185

Similarly a fantastic design was reported with the choice of imine 279, α -ketoacid 280 and isocyanide 281 [as](#page-22-0) [star](#page-40-0)ting materials, yielding Ugi adducts 282 which cyclizes to form 2,5 diketopiperazines 283 in the presence of TMSOTf in good yield (Scheme 79).¹⁸⁶

Dömling prepared the pentacyclic compound 285 by means of the combina[tion](#page-22-0) [of](#page-40-0) the Ugi and Pictet−Spengler reactions of tryptophan (Trp) 284, phthalic dialdehyde and tert-butyl isocyanide (Scheme 80).^{4a}

Figure 17. Polycyclic compounds via a two step Ugi condensation combined with the Pictet−Spengler reactions.

Scheme 79. Preparation of 2,5-Diketopiperazines

Scheme 80. Tryptophan in the Synthesis of Pentacyclic Derivative 285

The Müller group reported an impressive method to prepare highly substituted tetrahydro-β-carbolines 291 with five stereocenters utilizing a coupling-amination-aza-annulation-Pictet− Spengler MCR (Scheme 81).¹⁸⁷ In this one-pot reaction an acyl halide reacts with a terminal alkyne in the presence of palladium and copper to gi[ve](#page-40-0) the ynones 286. The later undergoes an aza-Michael addition of (R) -tryptamines 287 to form 288. The intermediate 290 forms form the condensation reaction of the indolyl amines 288 and α , β -unsaturated chlorides 289, followed by cyclization to the tetrahydro- β -carbolines 291 in 32−59% yield (Scheme 81). Similarly Gupta et al have prepared α-carboline derivatives via a three-component tandem reaction using acid chlorides, terminal alkynes, and 2-aminoindole hydrochlorides.188 Via N-acyliminium Pictet−Spengler reaction of tryptamine, cinnamaldehyde derivatives, and alkynoyl chlorides sequence b[y a](#page-40-0)n intramolecular Diels−Alder cycloaddition a diverse polycyclic alkaloid-like compounds were prepared.189a Another modification on the Pictet−Spengler

reaction is one-pot preparation of aldehydes came from alkenes and carbon monoxide (see Scheme $43)$.^{189b,c}

The synthesis of asymmetric indoloquinolizidines 295 from $β$ -ketoesters 292, $α, β$ -unsaturated [ald](#page-13-0)[ehyde](#page-40-0)s 293 and tryptamines was reported by Wu et al.¹⁹⁰ Both organocatalysts and prolinol ether 294 efficiently catalyzed this reaction to give 295 in moderate to excellent yields a[nd g](#page-40-0)ood to excellent enantioselectivities. The mechanism for the formation of the cyclic hemiacetal intermediate 296 appears to involve the conjugated addition of $β$ -ketoesters to $α, β$ -unsaturated aldehydes followed by treatment with tryptamine to yield the indoloquinolizidine 295 (Scheme 82). Alternatively similar products 295 were

Scheme 82. Michael and Pictet−Spengler Reaction in the Synthesis of Indoloquinolizidine

achieved by employing an asymmetric organocatalyzed one-pot three-component cascade reaction of tryptamines, alkyl propiolates, and α , β -unsaturated aldehydes.¹⁹¹

Chiral spirocyclic 2,6-dioxopiperazines A are prepared from normal amino acids as outlined in Scheme 83.192 Condensation of (L)-Trp-OMe, cyclohexanone and trimethylsilylcyanide leads to the α -amino nitrile 298 (Scheme 8[3\).](#page-23-0) [Th](#page-40-0)e Trp-derived amino nitrile 298 when treated with H_2SO_4 gives a new

Scheme 83. Chiral Spirocyclic 2,6-Dioxopiperazines 301

Scheme 84. Combination of Ugi and Diels−Alder Reaction for the Synthesis of 305

tetracyclic indoline derivative 299 (85%) and the carboxamide 300 (15%). Both 299 and 300 can be converted to 301. The indoline derivative 299 requires refluxing in 1 N HCl and the carboxamide 300 requires treatment with NaH to lead to the Trp-derived 2,6 dioxopiperazine 301. Further functionalization of the dioxopiperazine derivatives 301 to the corresponding derivatives A is achieved through alkylation of the latter with alkyl halides.

In another scenario Lu et al, via an sequential Ugi 4-CR, followed by an intramolecular Diels−Alder reaction, and oxidative aromatization made a wide diversity of benzofurans and indoles.¹⁹³ For instance 3-(1-tosyl-1H-indol-3-yl)acrylaldehyde 303 reacts with aniline, tert-butylisocyanide and phenylpropiolic acid producing the substituted indole derivative 302. Removal of the solvent (methanol) and stirring of the intermediate at 50 °C under nitrogen atmosphere for 4 h, after which DDQ was added as oxidant at room temperature, compound 305 was isolated in 73% (Scheme 84). As shown in Scheme 84, the Ugi intermediate 302 undergoes an intramolecular Diels−Alder reaction leading to 303. Product 305 is formed via a Hshift and subsequent oxidation reaction.

The Quinolino scaffold 307 and its derivatives were constructed in a combinatorial approach via the Ugi four-component reaction (U-4CR) and a Pd-catalyzed intramolecular arylation reaction (Scheme 85).¹⁹⁴ In this context, iodoindole 306, 4-chloroaniline, tert-butylisocyanide, and 2-methoxybenzaldehyde in methanol were c[ond](#page-24-0)[ense](#page-40-0)d and their Ugi-adduct was isolated in 88% yield which in the presence of palladium was cyclized to the quinolino-indole 307 in 92% yield. This method gives access to a large diversity of quinoline derivatives.

El Kaim and co-workers exposed the Ugi-adducts 308 to $Cu(OAc)$ ₂ in a novel method for the synthesis of spiroindolines 311 (Scheme 86).¹⁹⁵ This copper-catalyzed spirocyclization reaction probably proceeded via a radical mechanism starting with the gener[atio](#page-24-0)[n of](#page-40-0) enolates 309 (Scheme 86). This simple procedure produces final structures of remarkable complexity because of four points of molecular diversity [for](#page-24-0) the first step. This is a rare example of cyclization of a radical species which only gives one diastereomer. It may be explained either by a degradation of the isomer unable to cyclize with the pendant amide or by a ring-opening of the spiro iminium 310 leading to an epimerization of the peptidyl position.

Treatment of indoles with 5-hydroxyfuran-2(5H)-one 312 in the presence of catalyst 313 followed by addition of isocyanide and benzylamine give lactams 314 as a mixture of two diasteromers (Scheme 87).¹⁹⁶ 5-Hydroxyfuran-2(5H)-one 312 reacts with the diphenylprolinol silyl ether 313 to yield the activated intermediate [31](#page-24-0)5 [w](#page-40-0)hich undergoes Michael addition of indole in the beta position to form the acid aldehyde 316 in high enantioselectivity. The latter was precipitated successfully in an Ugi-condensation to yield lactams 314 in 75−95%.

Scheme 87. Combination of Michael Addition and Ugi Reaction

In a study toward the FeCl₃ mediated synthesis of α aminophosphonates, via a Kabachnic−Fields reaction between an aldehyde, amine and phosphite, the $bis(\alpha$ -aminophosphonate) 317 (68%) was isolated. In this specific case the condensation of terephthaldialdehyde, tryptamine and $P(OEt)_{3}$, leads to the formation of 317 (Figure 18).¹⁹⁷ In a biological investigation supported with docking studies, the indole derived 317 was shown to exhibit high activity on [the](#page-40-0) cell lines RAJI,

Figure 18. Bis(α -aminophosphonates) 317.

JURKAT and MCF-7. The observed activity was comparable to that of doxorubicin. N-alkylated α -methyltryptamine derivatives was generated by simple reductive amination of α -methyltryptamines with ketones using catalytic hydrogenation conditions (3 atm H_2 and 10% Pd on carbon).¹⁹⁸ This method was also applied to other primary and secondary amines using ketones and aldehydes.

In a one pot procedure N-boc protected 3-formyl indole 318 is first condensed with the aldol form of pyruvic acid and then cyclized with aniline derivatives to yield 2-indolyl-4-carboxylic acid quinolines 319 (Scheme 88). 199 It is certainly also possible

3532 dx.doi.org/10.1021/cr2003954 | Chem. Rev. 2012, 112, 3508−3549

that the indole-aldehyde first form the imine with aniline, followed by attack of the pyruvic acid aldol form, and subsequent cyclization. These compounds were then further functionalized and their antibacterial activities were evaluated on Gram-positive bacteria.

Gupta et al. have prepared a novel structure of pyrimido $[1,2$ a]-indoles 321 utilizing a three component reaction of functionalized indoles 320, alkynyl derivatives and acyl chlorides (Scheme 89). 200a The reaction must probably start with alkynone

Scheme 89. [Seq](#page-40-0)uential Sonogashira and [3 + 3]-Cyclocondensation Reactions

formation from acyl chloride and alkynes which efficiently catalyzed with palladium and cupper then produced alkynone and indole 320 undergoes cyclocondensation to form 321 in 63−82% yield (Scheme 89). Palladium also has catalyzed cyclization/ carboalkoxylation of alkenyl indoles.^{200b}

Mironov et al. reported the synthesis of 5-substituted indoles. Isoquinoline 322 attacks the alkene [32](#page-40-0)3 to yield the zwitterion 325, which subsequently cyclizes with indole-5-isocyanide 324 to give the 2,3-dihydro-10H-pyrrolo $[2,1-a]$ isoquinoline-1-one 326 in 76% yield (Scheme 90).²⁰¹

Van Leusen's imidazole approach involves the activation of tosylmethyl isocyanides 327 i[n w](#page-40-0)hich the acidic α -hydrogen is abstracted under these basic conditions to form 328. In situ condensation of the amine and aldehyde to an intermediate imine then facilitates the cyclization reaction yielding 329. Elimination of toluenesulphinic acid led to the imidazole 330 (Scheme 91).²⁰² When indole functionalized aldehydes or amines are used in this one pot approach, the corresponding imidazoles 331[−](#page-40-0)334 are efficiently obtained. The biological activities of some of these products were examined for instance against NCI-H460 and HCT-15 cancer cell lines gave good results (Scheme 91).²⁰³

3-Formylindoles react with glycine, potassium thiocyanate to give two related pr[odu](#page-40-0)cts, believed to form from different mechanistic pathways.²⁰⁴ N-methyl, N-n-butyl, or N-p-cyanoScheme 90. Pyrrolo[2,1-a]isoquinoline-1-ones 326 via 3-CR

phenoxypentyl-3-formylindole 335, glycine, and potassium thiocyanate in the presence of acetic anhydride form the gemdiacetylthio derivative 336 in 63−68% yield (Scheme 92). Acetic anhydride promotes the cyclization of 336 to 1-acetyl-2 thiohydantoin 337, with the generation of one equivalent [wate](#page-26-0)r molecule. Condensation of N-acetyl protected indole-aldehyde 338 with hydantoin 337 followed by partial N-deacetylation yields indolylmethylene 2-thiohydantoins 339 and 340 in 7% and 25% yield, respectively (Scheme 92).

Indium-mediated reaction of indole-3-carboxaldehydes, allyl bromide and electron-rich heteroarenes, [ele](#page-26-0)ctron-rich aromatics, or stabilized enols deliver a large diversity of interesting products such as unsymmetrical bisindolyl methanes 341 and 342, indolyl-uracil 343,²⁰⁵ indoly arylmethanes 344–347,²⁰⁶ and indolyl-pyrazoles 348 and indolyl-imidazoles 349 (Figure 19). 207

T[he](#page-40-0) reduction reaction of 3-alkyn[yl-in](#page-40-0)dole-2-carbaldehydes 350 in the presence of alkoxides, genera[ted](#page-26-0) [from](#page-40-0) alcohols and sodium metal, under mild reaction conditions leads to intramolecular cyclization and the isolation of the novel [1,4] oxazino $[4,3-\alpha]$ indole nucleus 351 (Scheme 93).²⁰⁸

Indole-3-carbaldehyde derivatives were used in the threecomponent aza-Diels−Alder reaction with [N-](#page-26-0)vi[nyl-](#page-40-0)2-pyrrolidinone 352 and aniline to give the corresponding products 353 (Scheme 94).²⁰⁹ Only 1-tosylindole-3-carbaldehyde proceeds well in this reaction with good yields while other indole-3 carbalde[hyd](#page-26-0)e[s su](#page-40-0)ch as indole-3-carbaldehyde and 1-methylindole-3-carbaldehyde did not give the desired products.

Scheme 91. Van Leu[sen](#page-40-0)'s Synthesis of Indolyl Imidazoles

Scheme 92. Hydantoin Derivatives of Indole

Figure 19. Allylic indolyl compounds.

Zhu et al. established a new synthetic protocol for the efficient and regiospecifc assembly of pyrido $[1,2-\alpha]$ indoles 354 or indolizines (Table 2). They employed indole-2-carbaldehyde or pyrrole-2-carbaldehyde, α -bromoketones, and alkynes in this novel protocol.²¹⁰ The mechanism of this reaction in basic media involves act[iva](#page-27-0)tion of the indole to N-alkylation by alkyl bromide and ca[rbo](#page-40-0)n between nitrogen and $R³$ as second nucleophilic attacks to β -position of eynone. The formed intermediate subsequently undergoes intramolecular cycloisomerization to afford the desired product 354.

Scheme 93. Nucleophilic Cyclization of δ-Acetylenic Aldehydes

Scheme 94. Indole-3-carbaldehydes in 3-C Aza-Diels−Alder Reactions

Majumder and Bhuyan have designed a novel reaction for the synthesis of α -carbolines 356 from indoles 355, which undergoes a Knoevenagel condensation with nitriles and then an intramolecular $\begin{bmatrix} 3 + 2 \end{bmatrix}$ -dipolar cycloaddition reaction by azides (Scheme 95). The last step consists of an aromatic nucleophilic substituted reaction.²¹¹

The N-boc-3-a[mid](#page-27-0)o indole 357 when treated with a range of aryl aldehydes and aromatic alk[yne](#page-40-0)s under acidic media and with microwave irradiation leads to the synthesis of a wide array of δ -carbolines 358 (Scheme 96).²¹² This reaction probably starts with boc-deprotection of 357 followed by imine formation with the aldehyde. Th[e in](#page-27-0) [situ](#page-40-0) produced indole imine acts as a heterodiene in a Diels−Alder cyclization reaction with the alkyne. Oxidation of the latter yields the δ-carbolines 358.

Huber et al. provided a facial method for the preparation of 3-substituted 4-cyano-1,2,3,4-tetrahydro-1-oxo-β-carbolines 360, by treating ethyl 3-(cyanomethyl)indole-2-carboxylate 359 with ammonia and addition of aldehydes or ketones to the reaction mixture (Scheme 97). 213 When cyclic ketones are used, the procedure allows a convenient synthesis of tetracyclic spiro compounds.

Substituted gramines 361, i[n](#page-27-0) [t](#page-27-0)he presence of tBuOK, react with acrolein and $β$ -keto-phosphonates, via a novel sequence involving a multicomponent condensation reaction followed by an anionic polycyclisation cascade, to afford the highly functionalized tetracyclic structures 362 (Scheme 98) in excellent overall yields and diastereoisomerically pure.²¹⁴ In this unique process, up to eight elementary transformation[s ta](#page-27-0)ke place with remarkably high chemo- and stereocontrol.

A new indolyl fullerenopyrrolidine 363 was synthesized with a three-component reaction of fullerene C_{60} , N-methylglycine, and indole-3-carbaldehyde in 40% yield (Scheme 99).²¹⁵ It is probably the aldehyde and amine which condensed to form an iminium salt followed by decarboxylation and a [su](#page-27-0)[bseq](#page-40-0)uent $[3 + 2]$ cycloaddition reaction to give 363. A similar reaction with ketones and chalcones has been reported.²¹⁶

Fang and co-workers made a series of tetracyclic compounds 366 bearing indole and thiophene rings. 217 Th[e ke](#page-40-0)y step is this three-component coupling of methyl thiophene-2-carboxylate with N-alkylindole-2-carbaldehyde an[d 4](#page-40-0)-methoxyacetophenone. The reaction is promoted by samarium diiodide (Scheme 100). This one-pot operation presumably proceeds

Table 2. Synthesis of Pyrido[1,2-a]indoles

Scheme 98. Preparation of Substituted Tetracycles 362

Scheme 99. C_{60} -Containing Indole

Scheme 96. Synthesis of δ -Carbolines

Scheme 97. β-Carbolines Synthesis via MCRs

Scheme 100. Carbazolothiophene-2-carboxylic Acid Derivatives 366

through the initial coupling of the ester with aldehyde to give a dienolate intermediate 364 which is then trapped by the acetophenone leading to compound 365. 9-Benzyl-4-methyl-4- (4-hydroxyphenyl)-10-oxo-4,10-dihydro-carbazolo[2,3-b]thiophene-2-carboxylic acid (366, when R^1 = benzyl and R^2 = H) show the most potent inhibition of the endothelin-1 induced increase of intracellular calcium ion concentration.

3,4-Dihydropyrimidine-2 (1H)-ones 367 form via the Biginelli three-component condensation of 3-formylindoles, urea or thiourea and ethyl acetoacetate. The antimicrobial activities of these compounds were evaluated (Scheme 101).²¹⁸ The reactions of

Scheme 101. Indolyl-pyrimidinone 367

benzocyclic α -ketoacids as carbonyl components in the Biginelli reaction have been investigated.²¹⁹

A series of indolyldihydropyridines 368−371 were synthesized via the Hantzsch reaction [of](#page-40-0) methyl acetyl acetate, 2- or 3-formylindoles under reflux conditions with methanol or ethanol and a solution of $30\% \text{ NH}_4\text{OH}$ (Figure 20). The

Figure 20. 2- and 3-Formylindoles in Hantzsch 1,4-Dihydropyridine Synthesis.

calcium channel blocking activity of these compound 368−371 were also evaluated.^{220a,b} One-pot transition-metal-catalyzed methylenation-hydroboration-Suzuki cross-coupling process with aldehydes has b[een r](#page-40-0)eported by Lebel et al.^{220c}

Surprisingly, the 3-CR of indane-1,3-dione 372, 5-amino-1,2 dihydropyrazol-3-one 373, and indole-3-carbalde[hyde](#page-40-0) results in

the isolation of tetracycle 374 in 44% yield instead of the expected indolyl derivative 375 (Scheme 102).²²¹ Compound 374 are also smoothly obtained in 34% yield when formaldehyde is used as the aldehyde. It appears t[hat](#page-40-0) oxidation of the indole subunit with oxygen occurs before oxidation of the dihydropyridine portion of the molecule resulting in C−C bond cleavage. A simple and efficient direct aldol reaction via the double activation of both aldehydes and ketones by ammonia to afford 2,2-dimethyl-6-aryl-4-pyrilidinones has been developed. 222

3-(Cyanoacetyl)indoles 376^{223} are prepared from indoles and cyanoacetic acid.²²⁴ These indoles were recently [us](#page-40-0)ed as starting substrates in the s[ynth](#page-40-0)esis of a variety of valuable azaheterocyclic comp[oun](#page-40-0)ds (Scheme 103). 2-Acylthiophene, 2- or 3-acylpyridine reacts with indole 376, aldehyde and NH4OAc and the 3-CR is cyclized [to](#page-29-0) the corresponding pyridine derivatives 377−379 under thermal conditions. The addition of DDQ improves the yield of the products (Scheme 103).225 A similar reaction under MW irradiation was investigated.²²⁶ Enamines, 376 and aldehydes in the Hantzsch [1,4-](#page-29-0)d[ihyd](#page-40-0)ropyridine synthesis was reported by Chen et al.²²⁷ Some [hete](#page-40-0)rocyclic scaffolds including 6-(2-furyl)-2- (1H-indol-3-yl)-4-arylpyridine-3-carbonitriles 380 are made utiliz[ing](#page-40-0) the same method (Scheme 103).²²⁸ Similarly the poly functional compound 381 is synthesized from cinnamil, compound 376 and ammonium acetat[e. Co](#page-29-0)[mp](#page-40-0)ound 382 also forms due to the condensation of terephthaldialdehyde, 2 acylpyridine, 376, and ammonium acetate at 120 °C or with microwave conditions (Scheme 103).^{228,229} Perumal extended the use of 3-cyanoacetyl indole 376 to the InCl₃ catalyzed preparation of pyridines derivatives [38](#page-29-0)3 [throug](#page-40-0)h four-component reactions of aldehydes, 376 and malononitrile in methanol (Scheme 103).²³⁰ Ji and co-workers described the efficient synthesis of 3-(2-furanyl)indole derivative 384 using a threecomponent re[acti](#page-40-0)on between 376, isocyanides and aromatic aldehydes [\(Sch](#page-29-0)eme 103).²³¹ 3-(Cyanoacetyl)indoles 376 also react with arylaldehydes and ammonium acetate under microwave irradiatio[n to](#page-29-0) [form](#page-40-0) 2,6-diindolyl pyridine derivatives 386 (Scheme 103).²³²

A series of polysubstituted (3-indolyl)pyrazolo[3,4-b]pyridines 385 and $(3'-indolyl)$ $(3'-indolyl)$ $(3'-indolyl)$ benzo $[h]$ quinoline derivatives 387 are synthesized from 376, aldehydes or ketones, 3-aminopyrazol or 1-aminonaphthylamine with microwave irradiation in good yield (Scheme 103).²³³ When ketones are used in this reaction, the corresponding 1,4-dihydropyridines are generated.²³⁴ 3-Pyranyl indo[les a](#page-29-0)r[e al](#page-40-0)so prepared in the same manner. These 3-pyranyl indoles were evaluated for antimicrobial, antioxidant, [and](#page-40-0) anticancer activities.^{235a,b} Compounds 376 were also employed in the synthesis of spiroindolyls.^{235c} Some of the compounds also

Scheme 103. 3-(Cyanoacetyl)indoles 376 in the Synthesis of Azaheterocyclic Compounds

showed good anticancer activity against MCF-7 breast cancer cell lines in comparison with Doxorubicin that was used as the control drug.

It was found that 2-vinylindoles 388 ($R^2 = H$) react with aldehydes and p-methoxyaniline to form tetrahydroquinoline derivatives 389 in good yield (Scheme 104).^{236a} On the other

Scheme 104. Chemoselective Three-comp[onen](#page-40-0)t Reaction of Aldehydes, p-Methoxyaniline and 2-Vinylindoles 288

hand when nitrogen protected analogues of 388 are used, tetrahydro-γ-carbolines 390 are isolated as sole product (Scheme 104). These regio- and chemoselective reactions are efficiently catalyzed with 3,5-dinitrobenzoic acid (DNBA) in 1,2-dichloroethane (DCE). Cobalt-catalyzed hydrohydrazination reaction of olefins has been reported.^{236b}

3-Acetylindole in the presence of lithium diisopropylamine (LDA) yields the dianion 291. The later reac[ts wi](#page-40-0)th benzophenone and subsequent addition of $Et_2C(COCl)_2$ results in the isolation of 1,3,5-triketooctene 292 in 18% yield (Scheme 105). 237 When an excess of LDA is used, regeneration of the enolate

anion appears to be possible after condensation of 291 with benzophenone. The carbon center of the enolate reacts with the dielectrophile, and subsequent ring closure involves attack at C-4 of the indole system, as attack at the nitrogen atom

Scheme 105. Reaction of the Dianion of 3-Acetylindole with Benzophenone and Diethylmalonic Dichloride

would lead to a highly strained product (Scheme 105). The exocyclic double bond $C=CPh_2$ is formed by elimination of water during the aqueous workup.

Perumal and his group have prepared 1,4-disubstituted 1,2,3 bis-triazoles 294. Products 294 were obtained from a variety of N-propargyl derivatives 293, benzyl bromides and sodium azide. CuI was employed as the catalyst in the presence of PEG-400 (Scheme 106).^{238a} These triazoles have also been screened for their biological activity. In addition the one-pot process reaction of

conjugated enyne, carbene complex and 2,3-dihydrofuran leading to polycyclic compounds have been reported.^{238b}

5. SYNTHESIS OF INDOLE BEARING [NAT](#page-41-0)URAL PRODUCTS VIA MCPS

Indole alkaloids usually have complex structures and chemists normally prefer to synthesize them by means of multistep processes. However, MCRs methodology offers an alternative method with a reduced number of steps.^{239a} For example, the schematic plan for the one-pot synthesis of compounds 295 and 297 is shown in Scheme 107. Th[ese p](#page-41-0)roducts are pre-

Scheme 107. MCRs in the Synthesis of (\pm) -allo-Yohimban Dihydrocorynantheine 304 296 and (\pm) -Nitraraine 298

cursors for (\pm) -allo-yohimban 296 and (\pm) -nitraraine 298, respectively (Scheme 107).^{239b,c} In this method the C=N bond of the cyclic imines, acryloyl chloride, and 2,4-pentadienyltin are coupled together, f[ollowe](#page-41-0)d by the spontaneous intramolecular Diels−Alder cycloaddition reaction (Scheme 107).

Tietze²⁴⁰ and Zhou have demonstrated that the biologically interesting hirsutine 303, an alkaloid, can be obtained with high selectivi[ty a](#page-41-0)nd efficiency from simple precursors such as 299 (when $R^1 = CO_2tBu$ and C3 has R configuration), Meldrum's acid and 4-methoxybenzyl butenyl ether (E: $Z \approx 1:1$). The

MCR consists of a sequence of domino reactions, namely a Knoevenagel−hetero-Diels−Alder reaction in the presence of ethylene diammonium diacetate (EDDA). Solvolysis of the formed lactone 301 with methanol/ K_2CO_3 and subsequent hydrogenation to 302 is followed by the condensation of the latter with methyl formate and treatment with diazomethane lead to the desired product 303 (Scheme 108).²⁴¹ Dihydrocorynantheine 304 is also analogously formed from 299 (when $R^1 = H$ and C3 is S configuration), Meldrum's a[cid](#page-41-0) and iso-propyl butenyl ether (Scheme 108).

Hirsutine 303 was successfully obtained by means of an alternative 3-component process in which dihydrocarboline 305 is treated with allyltributyltin and acryloyl chloride to furnish 306 in 75% yield. Cyclization of 306 in the presence of 4 mol % of Grubbs' catalyst 307, produced 308 in 87% yield. Several more steps are required for the ultimate preparation of Hirsutine 303 (Scheme 109).²⁴² Precursor 305 was also used in an enantioselective manner for the synthesis of azaeburnane analogue.^{243a} In situ p[repa](#page-31-0)r[atio](#page-41-0)n of 305 and participation in Ugi-condensation has been reported.^{243b}

Scheme 109. Grubbs' Catalyst 307 in Hirsutine 303 Preparation

Scheme 110. Synthesis of (−)-Dihydroantirhine 311

treated with $K_2CO_3/MeOH$ and a catalytic amount of Pd–C in methanol under a nitrogen atmosphere for 50 min. After that the mixture is stirred under a H_2 -atmosphere for 2 h at room temperature to give the benzoquinolizidine 315 via 313 and 314 with the correct stereochemistry at all stereogenic centers (Scheme 111). The alkaloid tubulosine 318 was successfully

Tietze et al. have prepared compound 309 via a 3-CR (Scheme 110). The latter is converted to 310 in the presence of H_2 catalyzed by Pd (Scheme 110).²⁴⁴ Reduction of the lactam 310 with LiAlH4 gives the indole alkaloid (−)-dihydroantirhine 311.

They also used this methodolog[y in](#page-41-0) the preparation of the lactone 312 (Scheme 111). 245 The cycloadduct 312 is directly

Scheme 111. Synthesis of [Tu](#page-41-0)bulosine 318

Scheme 112. Palladium Cross-Coupling in the Synthesis of YCHs

Scheme 113. Spergillamide 323 Formation via a Ugi Reaction

synthesized by reacting benzoquinolizidine 315 with O-benzyl seretonine 316 to form the amide 317; further transformations lead to 318.

Ishikura et al. employed a palladium catalyst for the one-pot carbonylation and cross-coupling reaction of indolylborates 319 with vinyl triflates 320 in the presence of carbon monoxide to prepare 2-acylindole 321. Compound 321 is converted to 322 by acid catalysis and then used as precursor for the synthesis of Yuehchukene (YCH)²⁴⁶ (Scheme 112).²⁴⁷ A highly efficient diastereoselective and enantioselective one-pot multistep reaction for the constructio[n o](#page-41-0)f cyclopenta[b[\]ind](#page-41-0)oles has been reported by Guo and co-workers.²⁴⁸

A typical natural products such as spergillamide 323 and its analogues are synthesized w[ith](#page-41-0) MCRs (Scheme 113).²⁴⁹ Dömling and co-workers reported a solution phase Ugi 4-CR method to get access to these compounds. The antibiotic [and](#page-41-0) cytotoxic activities of these compounds were measured. Several of the synthetic analogues are more potent than the original natural product.

Choshi et al. have prepared 324 and used it as a convenient starting material for the synthesis of grossularine-2 328 via a three-component cross coupling reaction (Scheme 114).²⁵⁰ The reaction of the triflate 324, carbon monoxide and Scheme 114. Synthesis of Grossularine-2

phenylboronic acid 325 is carried out at 80 °C in the presence of $PdCl_2(PPh_3)_2$ in anisole to provide the tetracyclic 2-benzoylpyrido $[2,3-b]$ -indole 326 (19%) along with 2-phenylpyrido[2,3-blindole 327 (58%). Subsequent hydrolysis of the N-SEM group of 326 with diluted acid gives a natural grossularine-2 (328) in 81% yield (Scheme 114).

A Mannich three-component reaction of hydroxyfuroindoline (−)-329, formaldehyde and silylenolether 330 as the key

Scheme 115. (+)-Madindolines A and B 331

Scheme 116. MW in the Construction of 4-Quinazoline-3,6 diones (332) Scaffold

step is used for the construction of (+)-Madindolines A and B 331 as potent IL-6 inhibitors (Scheme 115).²⁵¹

A microwave promoted 3-CR of o-amino benzoic acid, Dtryptophan methylester hydrochloride and relat[ed a](#page-41-0)mino acid leads to the corresponding natural products 332, glyantrypine $(R = H)$, fumiquinazoline F ($R = CH_3$) and fiscalin B ($R = i-Pr$) in overall yields of 55%, 39%, and 20%, respectively (Scheme 116).²⁵²

Recently, Takiguchi et al. reported the asymmetric total synthesis of two anticancer natural products 334 and 335 [em](#page-41-0)ploying a common tricyclic imine precursor 333 and an Ugi reaction (Scheme 117).²⁵³ Thus N-acetylardeemin was accessed by the Ugi-3CR of 333 with anthranilic acid, isocyanide, and N-protected D-Ala in toluene followed by deprotection and polycondensation, whereas fructigenine was synthesized by the Ugi-3CR of 333 with p-methoxybenzyl isocyanide and Boc-Phe with subsequent deprotection and diketopiperazine ring closure under basic conditions. In both cases, the Ugi reaction was highly stereoselective and the isocyanide attack takes place preferentially from the side opposite the bulky reverse-prenyl group of imine 333.

Kobayashi et al. have prepared tetracyclic chiral aminoacetals 339 via a one-pot procedure involving a Stille−Migita coupling, a 6π-azaelectrocyclization, and aminoacetal formation from vinylstannanes 338, vinyliodides 337, and cis-aminoindanol derivatives 336 (Scheme 118).²⁵⁴ The same group also reported a modified strategy for the successful synthesis of indole alkaloids (−)-20-epiuleine 340, ²⁵⁵ (−[\)-c](#page-41-0)orynantheidol 341, and (−)-corynantheidine 342 (Sc[hem](#page-34-0)e 118).²⁵⁶ In a combine vinylogous Mukaiyama−Mannic[h an](#page-41-0)d Diels−Alder reactions stereoselectively some hexahydroind[oles](#page-34-0) [were](#page-41-0) synthesized which are represent precursors of complex natural products.²⁵⁷

6. MISCELLANEOUS REACTIONS

Petasis et. al employed indolyl boronic acids, amines and aldehydes to open a new route for the highly selective functionalization of indoles.⁴⁷ The Petasis method was also used by Jiang et al. to prepare optically pure 3-indolyl N-substituted glycines 343 in 61 to 7[7%](#page-37-0). They employed a chiral amine as the chiral auxiliary in the absence of any catalyst (Scheme 119).²⁵⁸

In the course of a tandem alkylation/alkenylation reaction of 3-iodothiophene to construct trisubstituted thiophen[es v](#page-34-0)i[a a](#page-41-0) palladium-catalyzed system, Lautens and his group discovered that this methodology can also be extended to include indoles.²⁵⁹ After screening of the reaction parameters, compound 345 was isolated in 81% yield from 344 and tert-butyl acrylate [and](#page-41-0) 1-iodobutane (Scheme 120). It is worthy to note that the nature of the nitrogen protecting group is crucial for the efficiency of the reaction[. Th](#page-34-0)e use of a methyl protecting group yields only the direct Heck product.

Scheme 117. Ugi R[eac](#page-41-0)tion as a Key Step in the Asymmetric Total Syntheses of Two Pyrazino-pyrroloindole Alkaloids, Fructigenine A, and 5-N-Acetylardeemin

Scheme 119. Asymmetric Synthesis 3-Indolyl Substituted via Petasis Reaction

Scheme 120. Tandem Alkylation/Alkenylation of Indole 344

Müller and co-workers have developed a new method for the synthesis of naturally occurring indolyl pyrimidinones, such as meridianins and variolins via carbonylation/alkynylation of 3-iodoindoles and subsequent cyclocondensation.⁹² They have utilized a mixture of $(5\% \left[\text{Pd}(\text{PPh}_3)_2\text{Cl}_2\right])$ (E) and (1%) $[Pd(dppf)Cl₂]$ (F) as the catalytic system for efficient [co](#page-38-0)upling of 3-iodoindole, carbon monoxide and trimethylsilyl acetylene. The corresponding TMS-ethynylindoles 346−348 were isolated in good yields (Figure 21). But for the pyrrolopyridine 349 it turned out that a 0.05 equiv of $[\text{Pd}(PPh_3)_2\text{Cl}_2]$ provides a better yield than with the mixture of catalysts.

In a sequential reaction the protected indolecarboxaldehyde 350 is condensed with allyl amine to generate an intermediate imine that is treated with acetyl chloride and allylmagnesium bromide to furnish adducts 351 in a one pot process (Scheme 121).²⁶⁰ The latter is converted to the bridged bicyclic tetracycle 353 upon sequential ring closing metathesis (RCM) and Hec[k re](#page-35-0)a[ctio](#page-41-0)ns and treatment with ruthenium 352 and $Pd(OAc)$ ₂ respectively (Scheme 121). In another strategy the indolic aldehyde 350 is reacted with bis(trimethylsilyl)allylamine providing an intermediate imine t[hat i](#page-35-0)s treated in situ with acetyl chloride and the silyl ketene acetal 354 to furnish the amide 355 in 72% yield (Scheme 121).²⁶⁰ Bromoindole 355 is subjected to an intramolecular Heck cyclization reaction under microwave irradiation in the [prese](#page-35-0)[nce](#page-41-0) of $Pd(OAc)_{2}$, to give 356 as two isomers $\Delta^{4,5}$ and $\Delta^{3,4}$ in 85% and 12% yield, respectively.

Langer and co-workers started from N-methyl-2,3-dibromoindole 357 and utilized a Suzuki−Miyaura reaction with two different arylboronic acids. This one-pot approach is selective for the synthesis of unsymmetrical 2,3-diarylindoles 358 (Scheme 122)²⁶¹ Because position two of the indole ring is more electron-deficiencient than position three, the oxidative addition [of pa](#page-35-0)ll[adiu](#page-41-0)m to indole 357 occurs first at position two.

In a three component cyclization-anion transfer process the alkene 359 and stannyl indole 360 are undergoing concomitant carbonylation with carbon monoxide to produce 2-oxoindolyl derivatives 361 in good yield (Scheme 123).²⁶² Palladium catalyzed aminocarbonylation of aryl halides with an amine and carbon monoxide²⁶³ and carbonylatio[n of](#page-35-0) [ind](#page-41-0)olylborate with carbon monoxide and alkenes have been reported.²⁶⁴

Chataigner an[d P](#page-41-0)iettre have described the feasibility of a multicomponent domino $[4 + 2]/[3 + 2]$ cycloaddi[tion](#page-41-0) reaction with electron-poor nitroheteroaromatics.^{265a} 3-Nitroindole is used

Figure 21. Products via carbonylation/alkynylation of iodoindoles.

Scheme 121. Ru and Pd in the Formation of Tetracyclic 353 and Tricyclic 356, Respectively

Scheme 122. Synthesis of Unsymmetrical 2,3-Diarylindoles 358

Scheme 123. 2-Oxoindolyl Derivatives

for the rapid and efficient generation of tetracyclic dearomatized diamines 363 featuring a quaternary center at one of the ring junctions (Scheme 124). It is clear that it would be quite difficult to prepare these novel structures with alternative synthetic methods. The reduction products of 364 are considered to be potential candidates for the design and preparation of novel catalysts in the context of asymmetric synthesis. Activation of 3-nitroindole at high pressure allows it to behave as electron-poor heterodienes for a $[4 + 2]$ cyclization with vinyl ethers to form 362. The latter participates in a $[3 + 2]$ cycloaddition process with acrylate to give 363 in good yield. Already similar sequence cycloaddition has been reported for $β$ -nitrostyrenyl compounds.^{265b}

The Samarium-mediated reaction of indole 365 with carbonyl compounds followed by trap[ping](#page-41-0) of the intermediate 367 with allyl iodide leads to the synthesis of the highly stereoselective lactone 366 in 42−94% yield (Scheme 125).²⁶⁶

Scheme 124. Domino $[4 + 2]/[3 + 2]$ Cycloaddition Reaction

Scheme 125. Highly Stereoselective Substitution in the Dihydroindole Derivatives 366

Rhodium catalyzed nitrene addition to indolic derivatives has been reported.²⁶⁷ In the proposed mechanism for this reaction, amine and iodine(III) yield iminoiodane which is transferred to

Scheme 126. Access to 2,3-Substituted Indolines

[Ru^{III}] (Scheme 126). The produced metalonitrene is added to pi-bond of indole 368 to yield the aziridine intermediate 369. The latter undergoes fast ring-opening to give rise to a carbocation, the nature of which depends on the substitution pattern at the C3 position. In the case of 369 (R = H), formation of cation 370 is favored, whereas aziridine 369 $(R = Et)$ preferentially affords the more stable tertiary benzylic cation

373. Both intermediates (370 and 373) are then attacked in a cis manner by nucleophiles, presumably owing to conformational effects, however, aziridine 369 ($R = H$), in the presence of methanol as nucleophile gives the trans-product 372. A key feature of this transformation is the ability to control either its stereoselectivity by changing the nucleophile or its regioselectivity by the introduction of a substituent at C3.

7. CONCLUSIONS

It was demonstrated that the MCR technology has emerged as a valuable tool in the synthesis of natural and synthetic indole derivatives. This method offers the opportunity of synthesizing molecules via novel routes, which may be difficult or impossible with traditional methods. It also offers the possibility for the rapid synthesis of potential drugs without tedious and timeconsuming purification.

This technology depends mainly on the polycondensation of a large diversity of functional groups such as acids, isocyanides, amines, aldehydes, ketones, methylenes, aryls etc. or also a combination of two or three well-known reactions such as the Heck, Diels−Alder, Knoevenagel, Mannich, Ugi, Smile, Petasis reactions, etc.

In addition, research on MCRs opened exciting opportunities in the field of green organic reactions or green chemistry by minimizing waste, cost, and time. These are valuable factors in the chemical and pharmaceutical industries. The continued evolution of such methodologies promises new routes to readily synthesize complex molecules previously thought to inaccessible.

AUTHOR INFORMATION

Corresponding Author

*Fax: +98 21 88041344. E-mail: mshiri@alzahra.ac.ir.

Notes

The author declares no competi[ng](mailto:mshiri@alzahra.ac.ir) [financial](mailto:mshiri@alzahra.ac.ir) [interest.](mailto:mshiri@alzahra.ac.ir)

Biography

Morteza Shiri was born in 1978 at Hamedan, Iran. He obtained his B.Sc. in Applied Chemistry from Bu-Ali Sina University and a M.Sc in Organic Chemistry from Shiraz University with Prof. Habib Firouzabadi in 2004. He received his Ph.D under the supervision of Prof. Mohammad Ali Zolfigol in 2007 and then joined the Alzahra University. His research field involves the synthesis of heterocyclic derivatives, such as indoles and quinolines.

ACKNOWLEDGMENTS

The author thanks Alzahra University and Iran National Science Foundation (INSF) for financial support to our research group. Professor H. G. Kruger from UKZN in South

Africa is also greatly appreciated for proof-reading the manuscript and giving valuable suggestions.

DEDICATION

Dedicated to Prof. Mohammad Ali Zolfigol

REFERENCES

(1) (a) Sundberg, R. J. The Chemistry of Indoles; Academic Press: New York, 1970. (b) Sundberg, R. J. In Comprehensive Heterocyclic Chemistry; Katritzky, A. R., Rees, C. W., Eds.; Pergamon Press: Oxford, U.K., 1984; Vol. 4 (c) Joule, J. A. In Science of Synthesis, Houben−Weyl Methods of Molecular Transformations; Thomas, E. J., Ed.; George Thieme Verlag: Stuttgart, Germany, 2000; Vol. 10, Chapter 10.13 (d) Gribble, G. W. In Comprehensive Heterocyclic Chemistry II; Katritzky, A. R., Ress, C. W., Scriven, E. F. V., Bird, C. W., Eds.; Pergamon Press: Oxford, 1996; Vol. 2, p 207. (e) Sundberg, R. J. Indoles; Academic Press: London, 1996. (f) Gribble, G. W., In Top. Heterocycl. Chem., "Heterocyclic Scaffolds II: Reactions and Applications of Indoles" 2010, Vol. 26, Springer-Verlag, Berlin, Heidelberg.

(2) (a) Humphrey, G. R.; Kuethe, J. T. Chem. Rev. 2006, 106, 2875. (b) Cacchi, S.; Fabrizi, G. Chem. Rev. 2005, 105, 2873. (c) Battistuzzi, G.; Cacchi, S.; Fabrizi, G. Eur. J. Org. Chem. 2002, 2671. (d) Campo, J.; Garcia-Valverde, M.; Marcaccini, S.; Rojo, M. J.; Torroba, T. Org. Biomol. Chem. 2006, 4, 757. (e) Barluenga, J.; Rodreguez, F.; Fanans, F. J. Chem. Asian J. 2009, 4, 1036.

(3) (a) Shiri, M.; Zolfigol, M. A.; Kruger, G. H.; Tanbakouchian, Z. Chem. Rev. 2010, 110, 2250. (b) Bartoli, G.; Benciveni, G.; Dalpozzo, R. Chem. Soc. Rev. 2010, 39, 4449. (c) Bandini, M.; Eichholzer, A. Angew. Chem., Int. Ed. 2009, 48, 9608.

(4) (a) Dömling, A.; Ugi, I. Angew. Chem., Int. Ed. 2000, 39, 3168. (b) Dömling, A. Chem. Rev. 2006, 106, 17. (c) Zhu, J.; Bienaymé, H. Multicomponent Reactions; Wiley-VCH: Weinheim, Germany, 2005.

(d) Dömling, A.; Huang, Y. Synthesis 2010, 2859.

(5) Tsepalov, V. F. Zh. Fiz. Khim. 1961, 35, 1691.

(6) Strecker, A. Liebigs Ann. Chem. 1850, 75, 27.

(7) (a) Ugi, I.; Steinbrü ckner, C. DE-B 1 1959, 103, 337. (b)

(c) Ugi, I.; Meyr, R.; Fetzer, U. Angew. Chem., Int. Ed. 1959, 71, 386. (8) (a) Shaabani, A.; Maleki, A.; Rezayan, A.; Sarvary, A. Mol.

Diversity 2011, 15, 41. (b) Sadjadi, S.; Heravi, M. M. Tetrahedron 2011,

67, 2707. (c) Ivachtchenko, A. V.; Ivanenkov, Y. A.; Kysil, V. M.;

Krasavin, M. Y.; Ilyin, A. P. Russ. Chem. Rev. 2011, 79, 787.

(9) Sapi, J.; Laronze, J. Y. Arkivoc 2004, 208.

(10) Gerencser, J.; Dorman, G.; Darvas, F. QSAR Comb. Sci. 2006, 25, 439.

(11) (a) Oikawa, Y.; Hirasawa, H.; Yonemitsu, O. Tetrahedron Lett. 1978, 20, 1759. (b) Oikawa, Y.; Hirasawa, H.; Yonemitsu, O. Chem. Pharm. Bull. 1982, 30, 3092.

(12) Oikawa, Y.; Tanaka, M.; Hirasawa, H.; Yonemitsu, O. Chem. Pharm. Bull. 1981, 29, 1606.

(13) Liu, Z. X.; Ruan, X. X.; Huang, X. Chin. J. Chem. 2003, 21, 1497. (14) (a) Deb, M. L.; Bhuyan, P. J. Tetrahedron Lett. 2007, 48, 2159. (b) Wang, C.; Zhang, Y. Q.; Li, G. S.; Li, J. C.; Li, X. L. Chin. J. Org. Chem. 2003, 23, 1416. (c) Bailey, A. S.; Ellis, J. H.; Peach, J. M.; Pearman, M. L. J. Chem. Soc., Perkin Trans. 1 1983, 2425. (d) Cochard, F.; Sapi, J.; Laronze, J.-Y. Tetrahedron Lett. 2001, 42, 6291. (e) Cochard, F.; Laronze, M.; Prost, E.; Nuzillard, J.-M.; Auge, F.; ́ Petermann, C.; Sigaut, P.; Sapi, J.; Laronze, J.-Y. Eur. J. Org. Chem. 2002, 3481. (f) Jeannin, L.; Nagy, T. E.; Vassileva, E.; Sapi, J.; Laronze, J.-Y. Tetrahedron Lett. 1995, 36, 2057. (g) Nemes, C.; Laronze, J.-Y. Synthesis 1999, 254. (h) Jeannin, L.; Boisbrun, M.; Nemes, C.; Cochard, F.; Laronze, M.; Dardennes, E.; Kovacs-Kulyassa, A.; Sapi, J.; Laronze, J. Y. C. R. Chim. 2003, 6, 517. (i) Boisbrun, M.; Jeannin, L.; Toupet, L.; Laronze, J.-Y. Eur. J. Org. Chem. 2000, 3051. (j) Boisbrun, M.; Kovacs-Kulyassa, A.; Jeannin, L.; Sapi, J.; Toupet, L.; Laronze, J.-Y. Tetrahedron Lett. 2000, 41, 9771. (k) Dardennes, E.; Kovacs-Kulyassa, A.; Boisbrun, M.; Petermann, C.; Laronze, J. Y.; Sapi, J. Tetrahedron Asymmetry 2005, 16, 1329.

(15) Dardennes, E.; Kovacs-Kulyassa, A.; Renzetti, A.; Sapi, J.; Laronze, J. Y. Tetrahedron Lett. 2003, 44, 221.

(16) Nemes, C.; Jeannin, L.; Sapi, J.; Laronze, M.; Seghir, H.; Auge, F.; Laronze, J.-Y. Tetrahedron 2000, 56, 5479.

(17) Sabitha, G.; Kumar, R. M.; Reddy, M. S. K.; Yadav, J. S.; Rama Krishna, K. V. S.; Kunwar, A. C. Tetrahedron Lett. 2005, 46, 1659.

(18) Dardennes, E.; Gerard, S.; Petermann, C.; Sapi, J. Tetrahedron Asymmetry 2010, 21, 208.

(19) Deb, M. L.; Bhuyan, P. J. Synthesis 2008, 2891.

(20) Mahadevan, A.; Sard, H.; Gonzalez, M.; McKew, J. C. Tetrahedron Lett. 2003, 44, 4589.

(21) Duan, Y.; Chen, M.-W.; Ye, Z.-S.; Wang, D.-S.; Chen, Q.-A.; Zhou, Y.-G. Chem.-Eur. J. 2011, 17, 7193.

(22) (a) Cao, L. L.; Wang, D. S.; Jiang, G. F.; Zhou, Y. G. Tetrahedron Lett. 2011, 52, 2837. (b) Duan, Y.; Chen, M.-W.; Chen,

Q.-A.; Yu, C.-B.; Zhou, Y.-G. Org. Biomol. Chem. 2012, 10, 1235.

(23) Guo, X.; Pan, S.; Liu, J.; Li, Z. J. Org. Chem. 2009, 74, 8848. (24) Li, M.; Yang, J.; Gu, Y. Chem. Commun. 2011, 353, 1551.

(25) Renzetti, A.; Dardennes, E.; Fontana, A.; De Maria, P.; Sapi, J.; Gerard, S. J. Org. Chem. 2008, 73, 6824.

(26) Epifano, F.; Genovese, S.; Rosati, O.; Tagliapietra, S.; Pelucchini, C.; Curini, M. Tetrahedron Lett. 2011, 52, 568.

(27) (a) Gu, Y.; Barrault, J.; Jerome, F. Adv. Synth. Catal. 2009, 351, 3269. (b) Gu, Y.; De Sousa, R.; Frapper, G.; Bachmann, C.; Barrault,

J.; Jerome, F. Green Chem. 2009, 11, 1968.

(28) (a) Sui, Y.; Liu, L.; Zhao, J. L.; Wang, D.; Chen, Y. J. Tetrahedron Lett. 2007, 48, 3779. (b) Ballini, R.; Clemente, R. R.; Palmieri, A.; Petrini, M. Adv. Synth. Catal. 2006, 348, 191.

(29) Gerard, S.; Renzetti, A.; Lefevre, B.; Fontana, A.; de Maria, P.; Sapi, J. Tetrahedron 2010, 66, 3065.

(30) Appendino, G.; Cicione, L.; Minassi, A. Tetrahedron Lett. 2009, 50, 5559.

(31) (a) Yamamoto, Y.; Harimaya, K. Chem. Lett. 2004, 33, 238. (b) Yamamoto, Y.; Kurazono, M. Bioorg. Med. Chem. Lett. 2007, 17, 1626.

(32) Tan, J.-N.; Li, M.; Gu, Y. Green Chem. 2010, 12, 908.

(33) Cochard, F.; Laronze, M.; Sigaut, P.; Sapi, J.; Laronze, J. Y. Tetrahedron Lett. 2004, 45, 1703.

(34) Laronze-Cochard, M.; Cochard, F.; Daras, E.; Lansiaux, A.; Brassart, B.; Vanquelef, E.; Prost, E.; Nuzillard, J.-M.; Baldeyrou, B.; Goosens, J.-F.; Lozach, O.; Meijer, L.; Riou, J.-F.; Henon, E.; Sapi, J. Org. Biomol. Chem. 2010, 8, 4625.

(35) Shanthi, G.; Perumal, P. T. Tetrahedron Lett. 2007, 48, 6785.

(36) Shanthi, G.; Perumal, P. T.; Rao, U.; Sehgal, P. K. Indian J. Chem. 2009, 48B, 1319.

(37) Qu, Y.; Ke, F.; Zhou, L.; Li, Z.; Xiang, H.; Wuab, D.; Zhou, X. Chem. Commun. 2011, 47, 3912.

(38) Chen, W.; Cai, Y.; Fu, X.; Liu, X.; Lin, L.; Feng, X. Org. Lett. 2011, 13, 4910.

(39) Shen, Z. L.; Ji, S. J.; Wang, S. Y.; Zeng, X. F. Tetrahedron 2005, 61, 10552.

(40) Palmieri, A.; Petrini, M. J. Org. Chem. 2007, 72, 1863.

(41) Adamo, M. F. A.; Konda, V. R. Org. Lett. 2007, 9, 303.

(42) El Kaim, L.; Grimaud, L.; Oble, J. Org. Biomol. Chem. 2006, 4, 3410.

(43) Terada, M.; Machioka, K.; Sorimachi, K. Angew. Chem., Int. Ed. 2009, 48, 2553.

(44) Kumar, A.; Sharma, S.; Maurya, R. A. Tetrahedron Lett. 2009, 50, 5937.

(45) Wang, M.-Z.; Zhou, C.-Y.; Wong, M.-K.; Che, C.-M. Chem. Eur. J. 2010, 16, 5723.

(46) Liu, J.; He, T.; Wang, L. Tetrahedron 2011, 67, 3420.

(47) Petasis, N. A.; Goodman, A.; Zavialov, I. A. Tetrahedron 1997, 53, 16463.

(48) Naskar, D.; Neogi, S.; Roy, A.; Mandal, A. B. Tetrahedron Lett. 2008, 49, 6762.

(49) Royer, D.; Wong, Y. S.; Ple, S.; Chiaroni, A.; Diker, K.; Levy, J. Tetrahedron 2008, 64, 9607.

(50) (a) Diker, K.; Doe de Maindreville, M.; Royer, D.; Le Provost, F.; Levy, J. Tetrahedron Lett. 1999, 40, 7463. (b) Diker, K.; Doe de Maindreville, M.; Levy, J. Tetrahedron Lett. 1999, 40, 7459.

(51) Laudet, B.; Moucadel, V.; Prudent, R.; Filhol, O.; Wong, Y. S.; Royer, D.; Cochet, C. Mol. Cell. Biochem. 2008, 316, 63.

(52) Noland, W. E.; Wahstrom, M. J.; Konkel, M. J.; Brigham, M. E.; Trowbridge, A. G.; Konkel, L. M.; Gourneau, R. P.; Scholten, C. A.; Lee, N. H.; Condoluci, J. J.; Gac, T. S.; Pour, M. M.; Radford, P. M. J. Heterocyclic Chem. 1993, 30, 81.

(53) Ty, N.; Dupeyre, G.; Chabot, G. G.; Seguin, J.; Quentin, L.; Chiaroni, A.; Tillequin, F.; Scherman, D.; Michel, S.; Cachet, X. Eur. J. Med. Chem. 2010, 45, 3726.

- (54) Shiri, M.; Zolfigol, M. A. Tetrahedron 2009, 65, 587.
- (55) Shirakawa, S.; Kobayashi, S. Org. Lett. 2006, 8, 4939.
- (56) Olyaei, A.; Shams, B.; Sadeghpour, M.; Gesmati, F.; Razaziane, Z. Tetrahedron Lett. 2010, 51, 6086.
- (57) Srihari, P.; Singh, V. K.; Bhunia, D. C.; Yadav, J. S. Tetrahedron Lett. 2009, 50, 3763.
- (58) Dai, H.-G.; Li, J.-T.; Li, T.-S. Synth. Commun. 2006, 36, 1829. (59) Sharifi, A.; Mirzaei, M.; Naimi-Jamal, M. R. Monatsh. Chem. 2001, 132, 875.

(60) El-Gendy, A. A.; Said, M. M.; Ghareb, N.; Mostafa, Y. M.; El-Ashry, E. S. H. Arch. Pharm. Chem. Life Sci. 2008, 341, 294.

- (61) Kundu, D.; Bagdi, A. K.; Majee, A.; Hajra, A. Synlett 2011, 1165. (62) Rao, V. K.; Chhikara, B. S.; Shirazi, A. N.; Tiwari, R.; Parang, K.; Kumar, A. Bioorg. Med. Chem. Lett. 2011, 21, 3511.
- (63) (a) Yadav, D. K.; Patel, R.; Srivastava, V. P.; Watal, G.; Yadav, L. D. S. Tetrahedron Lett. 2010, 51, 5701. (b) Kumar, A.; Gupta, M.;
- Kumar, K. M. Green Chem. 2012, 14, 290.

(64) Das, B.; Kumar, J. N.; Kumar, A. S.; Damodar, K. Synthesis 2010, 914.

- (65) Matsumoto, K.; Uchida, T.; Hashimoto, S.; Yonezawa, Y.; Iida, H.; Kakehi, A.; Otani, S. Heterocycles 1993, 36, 2215.
- (66) Saidi, M. R.; Azizi, N.; Naimi-Jamal, M. R. Tetrahedron Lett. 2001, 42, 8111.
- (67) Zhang, H.-C.; Brumfield, K. K.; Jaroskova, L.; Maryanoff, B. E. Tetrahedron Lett. 1998, 39, 4449.
- (68) McFarland, J. M.; Joshi, N. S.; Francis, M. B. J. Am. Chem. Soc. 2008, 130, 7639.
- (69) Scholtz, M. Ber. 1913, 46, 2138.
- (70) (a) Passerini, M.; Bonciani, M. Gazz. Chim. Ital. 1933, 63, 138.
- (b) Passerini, M.; Albani, F. Gazz. Chim. Ital. 1935, 65, 933.
- (71) Janczuk, A.; Zhang, W.; Xie, W.; Lou, S.; Cheng, J. P.; Wang, P. G. Tetrahedron Lett. 2002, 43, 4271.
- (72) Ghandi, M.; Taheri, A. Molecules 2009, 14, 1056.
- (73) Zhao, J. L.; Liu, L.; Zhang, H. B.; Wu, Y. C.; Wang, D.; Chen, Y. J. Synlett 2006, 96.
- (74) Prajapati, D.; Gadhwal, S.; Sarma, R. Lett. Org. Chem. 2008, 5, 365.
- (75) Zhang, G. W.; Wang, L.; Nie, J.; Ma, J. A. Adv. Synth. Catal. 2008, 350, 1457.
- (76) Kang, Q.; Zhao, Z.-A.; You, S.-L. Tetrahedron 2009, 65, 1603.
- (77) Desimoni, G.; Faita, G.; Mella, M.; Toscanini, M.; Boiocchi, M. Eur. J. Org. Chem. 2008, 6232.
- (78) Desimoni, G.; Faita, G.; Mella, M.; Toscanini, M.; Boiocchi, M. Eur. J. Org. Chem. 2009, 2627.
- (79) Chaubet, G.; Maillard, L. T.; Martinez, J.; Masurier, N. Tetrahedron 2011, 67, 4897.
- (80) (a) Lavilla, R.; Bernabeu, M. C.; Carranco, I.; Diaz, J. L. Org. Lett. 2003, 5, 717. (b) Carranco, I.; Díaz, J. L; Jiménez, O.; Vendrell,
- M.; Albericio, F.; Royo, M.; Lavilla, R. J. Comb. Chem. 2005, 7, 33. (81) Sarkar, N.; Banerjee, A.; Nelson, S. G. J. Am. Chem. Soc. 2008, 130, 9222.
- (82) (a) Yu, X. X.; Wu, J. J. Comb. Chem. 2010, 12, 238. (b) Markina, N. A.; Mancuso, R.; Neuenswander, B.; Lushington, G. H.; Larock, R. C. ACS Comb. Sci. 2011, 13, 265. (c) Barluenga, J.; V1zquez-Villa, H.; Merino, I.; Ballesteros, A.; Gonz1lez, J. M. Chem. Eur. J. 2006, 12, 5790.
- (83) Qiu, G.; Ding, Q.; Peng, Y.; Wub, J. Tetrahedron Lett. 2010, 51, 4391.
- (84) Yadav, J. S.; Reddy, B. V. S.; Yadav, N. N.; Gupta, M. K. Tetrahedron Lett. 2008, 49, 2815.
- (85) (a) Nassiri, M.; Maghsoodlou, M. T.; Heydari, R.; Habibi Khorassani, S. M. Mol. Diversity 2008, 12, 111. (b) Abbasinejad, M. A.; Mosslemin, M. H.; Fazlinia, A.; Esfandiari, H. Synth. Commun. 2010, 40, 385.
- (86) Lavilla, R.; Gotsens, T.; Guerrero, M.; Bosch, J. Synthesis 1995, 382.
- (87) Sheinkman, A. K.; Lopatinskaya, K. Y.; Klyuev, N. A.; Torosyan, Z. K. Chem. Heterocyl. Compd. 1980, 16, 174.
- (88) Corey, E. J.; Tian, Y. Org. Lett. 2005, 7, 5535.
- (89) Fokas, D.; Yu, L.; Baldino, C. M. Mol. Diversity 2005, 9, 81.
- (90) Fokas, D.; Hamzik, J. A. Synlett 2009, 581.
- (91) Merkul, E.; Oeser, T.; Müller, T. J. J. Chem.-Eur. J. 2009, 15, 5006.
- (92) Karpov, A. S.; Merkul, E.; Rominger, F.; Müller, T. J. J. Angew. Chem., Int. Ed. 2005, 44, 6951.
- (93) Cernak, T. A.; Lambert, T. H. J. Am. Chem. Soc. 2009, 131, 3124.
- (94) Lang, R.; Wu, J.; Shi, L.; Xia, C.; Li, F. Chem. Commun. 2011, 47, 12553.
- (95) Huang, Y.; Walji, A. M.; Larsen, C. H.; MacMillan, D. W. C. J. Am. Chem. Soc. 2005, 127, 15051.
- (96) Chi, Y.; Scroggins, S. T.; Frechet, J. M. J. J. Am. Chem. Soc. 2008, 130, 6322.
- (97) Galzerano, P.; Pesciaioli, F.; Mazzanti, A.; Bartoli, G.; Melchiorre, P. Angew. Chem., Int. Ed. 2009, 48, 7892.
- (98) Arai, T.; Yokoyama, N. Angew. Chem., Int. Ed. 2008, 47, 4989. (99) Arai, T.; Wasai, M.; Yokoyama, N. J. Org. Chem. 2011, 76, 2909. (100) Enders, D.; Wang, C.; Mukanova, M.; Greb, A. Chem. Commun. 2010, 46, 2447.
- (101) (a) Itoh, T.; Uehara, H.; Ogiso, K.; Nomura, S.; Hayase, S.; Kawatsura, M. Chem. Lett. 2007, 36, 50. (b) Kobayashi, J. K.; Matsui, S. I.; Ogiso, K.; Nomura, S.; Hayase, S.; Kawatsura, M.; Itoh, T. Tetrahedron 2010, 66, 3917.
- (102) (a) Bandini, M.; Cozzi, P. G.; Giacomini, M.; Melchiorre, P.; Selva, S.; Umani-Ronchi, A. J. Org. Chem. 2002, 67, 3700. (b) Kantam, M.; Laha, L.; Yadav, S. J.; Choudary, B. M.; Sreedhar, B. Adv. Synth. Catal. 2006, 348, 867.
- (103) Zhao, X. D.; Yu, Z. K.; Xu, T. Y.; Wu, P.; Yu, H. F. Org. Lett. 2007, 9, 5263.
- (104) (a) Yavari, I.; Hossaini, Z.; Karimi, E. Monatsh. Chem. 2007, 138, 1267. (b) Anary-Abbasinejad, M.; Rostami, N.; Parhami, A.; Hassanabadi, A. J. Chem. Res. (S) 2007, 291.
- (105) Anary-Abbasinejad, M.; Mosslemin, M. H.; Anaraki-Ardakani, H.; Tahan, S. J. Chem. Res. (S) 2006, 306.
- (106) Tsuchimoto, T.; Kanbara, M. Org. Lett. 2011, 13, 912.
- (107) Wang, J.; Wang, J.; Zhu, Y.; Lu, P.; Wang, Y. Chem. Commun. 2011, 47, 3275.
- (108) Sharma, S. K.; Gupta, S.; Saifuddin, M.; Mandadapu, A. K.; Agarwal, P. K.; Gauniyal, H. M.; Kundu, B. Tetrahedron Lett. 2011, 52, 65.
- (109) Sharma, S. K.; Mandadapu, A. K.; Kumar, B.; Kundu, B. J. Org. Chem. 2011, 76, 6798.
- (110) Facoetti, D.; Abbiati, G.; Dell'Acqua, M.; Rossi, E. Tetrahedron 2011, 67, 6833.
- (111) Yavari, I.; Habibi, A. Synthesis 2004, 989.
- (112) Ren, X.; Chen, J.; Chen, F.; Cheng, J. Chem. Commun. 2011, 47, 6725.
- (113) Lin, R.; Ding, S.; Shi, Z.; Jiao, N. Org. Lett. 2011, 13, 4498.
- (114) (a) Simoneau, C. A.; Ganem, B. Nature Protocols 2008, 3, 1249.
- (b) Simoneau, C. A.; Ganem, B. Tetrahedron 2005, 61, 11374.
- (115) Wagaw, S.; Yang, B. H.; Buchwald, S. L. J. Am. Chem. Soc. 1999, 121, 10251.
- (116) (a) KIhling, P.; Schmidt, A. M.; Eilbracht, P. Org. Lett. 2003, 5, 3213. (b) Mihovilovic, M. D.; Stanetty, P. Angew. Chem., Int. Ed. 2007, 46, 3612. (c) Schmidt, A. M.; Eilbracht, P. Org. Biomol. Chem. 2005, 3, 2333. (d) Schmidt, A. M.; Eilbracht, P. J. Org. Chem. 2005, 70, 5528. (e) Angelovski, G.; Keranen, M. D.; Linnepe, P.; Grudzielanek, S.; ̈
- Eilbracht, P. Adv. Synth. Catal. 2006, 348, 1193. (f) Bondzic, B. P.; Farwick, A.; Liebich, J.; Eilbracht, P. Org. Biomol. Chem. 2008, 6, 3723.
- (117) Linnepe, P.; Schmidt, A. M.; Eilbracht, P. Org. Biomol. Chem. 2006, 4, 302.
- (118) Khedkar, V.; Tillack, A.; Michalik, M.; Beller, M. Tetrahedron Lett. 2004, 45, 3123.
- (119) El Kaïm, L.; Grimaud, L.; Ronsseray, C. Synlett 2010, 2296.
- (120) Yeo, S. J.; Liu, Y.; Wang, X. Tetrahedron 2012, 68, 813.
- (121) Grotkopp, O.; Ahmad, A.; Frank, W.; Müller, T. J. J. Org. Biomol. Chem. 2011, 9, 8130.
- (122) Borthakur, M.; Gogoi, S.; Gogoi, J.; Boruah, R. C. Tetrahedron Lett. 2010, 51, 5160.
- (123) Jiang, B.; Yi, M.-S.; Shi, F.; Tu, S.-J.; Pindi, S.; McDowell, P.; Li, G. Chem. Commun. 2012, 48, 808.
- (124) Onitsuka, K.; Suzuki, S.; Takahashi, S. Tetrahedron Lett. 2002, 43, 6197.
- (125) Kamijo, S.; Yamamoto, Y. J. Am. Chem. Soc. 2002, 124, 11940. (126) Gabriele, B.; Veltri, L.; Salerno, G.; Mancuso, R.; Costa, M. Adv. Synth. Catal. 2010, 352, 3355.
- (127) (a) Marchetti, M.; Paganelli, S.; Carboni, D.; Ulgheri, F.; Del Ponte, G. J. Mol. Catal. A 2008, 288, 103. (b) Iwasaki, M.; Kobayashi, Y.; Li, J.-P.; Matsuzaka, H.; Ishii, Y.; Hidai, M. J. Org. Chem. 1991, 56,
- 1922. (128) (a) Kreye, O.; Westermann, B.; Wessjohann, L. A. Synlett 2007, 3188. (b) Rubinshtein, M.; James, C. R.; Young, J. L.; Ma, Y. J.;
- Kobayashi, Y.; Gianneschi, N. C.; Yang, J. Org. Lett. 2010, 12, 3560. (c) Isaacson, J.; Gilley, C.; Kobayashi, Y. J. Org. Chem. 2007, 72, 3913.
- (d) Vamos, M.; Ozboya, K.; Kobayashi, Y. Synlett 2007, 1595.
- (129) Gilley, C. B.; Buller, M. J.; Kobayashi, Y. Org. Lett. 2007, 9, 3631.
- (130) Rhoden, C. R. B.; Rivera, D. G.; Kreye, O.; Bauer, A. K.; Westermann, B.; Wessjohann, L. A. J. Comb. Chem. 2009, 11, 1078.
- (131) Yan, C. G.; Wang, Q. F.; Song, X. K.; Sun, J. J. Org. Chem. 2009, 74, 710.
- (132) Shestopalov, A. M.; Naumov, O. A.; Nesterov, V. N. Russ. Chem. Bull. 2003, 52, 179.
- (133) Damavandi, S. Heterocycl. Commun. 2011, 17, 125.
- (134) Schneekloth, J. S.; Kim, J. J.; Sorensen, E. J. Tetrahedron 2009, 65, 3096.
- (135) Deyrup, J. A.; Vestling, M. M.; Hagan, W. V.; Yun, H. Y. Tetrahedron 1969, 25, 1467.
- (136) Mossetti, R.; Pirali, T.; Saggiorato, D.; Tron, G. C. Chem. Commun. 2011, 47, 6966.
- (137) Chernyak, D.; Chernyak, N.; Gevorgyan, V. Adv. Synth. Catal. 2010, 352, 961.
- (138) Kalinski, C.; Umkehrer, M.; Schmidt, J.; Ross, G.; Kolb, J.; Burdack, C.; Hiller, W.; Hoffmann, S. D. Tetrahedron Lett. 2006, 47, 4683.
- (139) Coutts, I. G. C.; Southcott, M. R. J. Chem. Soc., Perkin Trans. I 1990, 767.
- (140) Heck, R. F.; Nolley, J. P. J. Org. Chem. 1972, 37, 2320.
- (141) (a) El Kaim, L.; Gizzi, M.; Grimaud, L. Org. Lett. 2008, 10, 3417. (b) El Kaim, L.; Grimaud, L. Mol. Diversity 2009, 1.
- (142) (a) Ackermann, L. Org. Lett. 2005, 7, 439. (b) Kaspar, L. T.; Ackermann, L. Tetrahedron 2005, 61, 11311. (c) Tang, Z.-Y.; Hu, Q.-S.
- Adv. Synth. Catal. 2006, 348, 846.
- (143) Barluenga, J.; Jimenez-Aquino, A.; Valdes, C.; Aznar, F. Angew. Chem., Int. Ed. 2007, 46, 1529.
- (144) Zhu, X.; Xu, X.-P.; Sun, C.; Chen, T.; Shen, Z.-L.; Ji, S.-J. Tetrahedron 2011, 67, 6375.
- (145) Arcadi, A.; Cacchi, S.; Carnicelli, V.; MarinelIi, F. Tetrahedron 1994, 50, 437.
- (146) Cacchi, S.; Fabrizi, G.; Parisi, L. M. Synthesis 2004, 1889.
- (147) Cacchi, S.; Fabrizi, G.; Pace, P.; Marinelli, F. Synlett 1999, 620. (148) Battistuzzi, G.; Cacchi, S.; Fabrizi, G.; Marinelli, F.; Parisi,
- L. M. Org. Lett. 2002, 4, 1355. (149) Haubrich, D. R.; Ward, S. J.; Baizman, E.; Bell, M. R.; Bradford,
- J.; Ferrari, R.; Miller, M.; Perrone, M.; Pierson, A. K.; Saelens, J. K.; Luttinger, D. J. Pharmacol. Exp. Ther. 1990, 255, 511.
- (150) Arthuis, M.; Pontikis, R.; Florent, J. C. Org. Lett. 2009, 11, 4608.
- (151) Vieira, T. O.; Meaney, L. A.; Shi, Y.-L.; Alper, H. Org. Lett. 2008, 10, 4899.
- (152) Ohno, H.; Ohta, Y.; Oishi, S.; Fujii, N. Angew. Chem., Int. Ed. 2007, 46, 2295.
- (153) Ohta, Y.; Chiba, H.; Oishi, S.; Fujii, N.; Ohno, H. Org. Lett. 2008, 10, 3535.
- (154) Ohta, Y.; Oishi, S.; Fujii, N.; Ohno, H. Org. Lett. 2009, 11, 1979.
- (155) Ohta, Y.; Chiba, H.; Oishi, S.; Fujii, N.; Ohno, H. J. Org. Chem. 2009, 74, 7052.
- (156) Zhang, X.; Corma, A. Angew. Chem., Int. Ed. 2008, 47, 4430.
- (157) (a) Olivi, N.; Spruyt, P.; Peyrat, J. F.; Alami, M.; Brion, J. D. Tetrahedron Lett. 2004, 45, 2607. (b) Russo, O.; Messaoudi, S.; Hamze, A.; Olivi, N.; Peyrat, J. F.; Brion, J. D.; Sicsic, S.; Berque-Bestel, I.; Alami, M. Tetrahedron 2007, 63, 10671.
- (158) (a) Suzuki, Y.; Ohta, Y.; Oishi, S.; Fujii, N.; Ohno, H. J. Org. Chem. 2009, 74, 4246. (b) Ohta, Y. Yakugaku Zasshi 2010, 130, 925.
- (159) (a) Chaplin, J. H.; Flynn, B. L. Chem. Commun. 2001, 1594. (b) Flynn, B. L.; Hamel, E.; Jung, M. K. J. Med. Chem. 2002, 45, 2670.
- (160) Sonogashira, K.; Tohda, Y.; Hagihara, N. Tetrahedron Lett. 1975, 16, 4467.
- (161) (a) Lu, B. Z.; Zhao, W. Y.; Wei, H. X.; Dufour, M.; Farina, V.; Senanayake, C. H. Org. Lett. 2006, 8, 3271. (b) Chen, Y.; Markina, N. A.; Larock, R. C. Tetrahedron 2009, 65, 8908.
- (162) Rao, R. M.; Reddy, U.; Alinakhi, C. H.; Mulakayala, N.; Alvala, M.; Arunasree, M. K.; Poondra, R. R.; Iqbal, J.; Pal, M. Org. Biomol. Chem. 2011, 9, 3808.
- (163) Yu, M. S.; de Leon, L. L.; McGuire, M. A.; Botha, G. Tetrahedron Lett. 1998, 39, 9347.
- (164) Leogane, O.; Lebel, H. Angew. Chem., Int. Ed. 2008, 47, 350. (165) Chen, Z.; Zheng, D.; Wu, J. Org. Lett. 2011, 13, 848.
- (166) Willis, M. C.; Brace, G. N.; Findlay, T. J. K.; Holmes, I. P. Adv. Synth. Catal. 2006, 348, 851.
- (167) Yao, P.-Y.; Zhang, Y.; Hsung, R. P.; Zhao, K. Org. Lett. 2008, 10, 4275.
- (168) Mizoguchi, H.; Oguri, H.; Tsuge, K.; Oikawa, H. Org. Lett. 2009, 11, 3016.
- (169) (a) Thormann, M.; Almstetter, M. Int. Patent WO 2004/ 099124, 2004. (b) Thormann, M.; Almstetter, M. U.S. Patent US 2007/0060624, 2007.
- (170) Waller, R. W.; Diorazio, L. J.; Taylor, B. A.; Motherwell, W. B.; Sheppard, T. D. Tetrahedron 2010, 66, 6496.
- (171) (a) Waki, M.; Minematsu, Y.; Meienhofer, J.; Izumiya, N. Chem. Lett. 1979, 823. (b) Waki, M.; Meienhofer, J. J. Am. Chem. Soc. 1977, 99, 6075.
- (172) Shiri, M.; Zolfigol, M. A.; Rastegar-Faal, T. Unpublished results.
- (173) Hulme, C.; Dietrich, J. Mol. Diversity 2009, 13, 195.
- (174) Ilyn, A. P.; Kuzovkova, J. A.; Potapov, V. V.; Shkirando, A. M.; Kovrigin, D. I.; Tkachenko, S. E.; Ivachtchenko, A. V. Tetrahedron Lett. 2005, 46, 881.
- (175) Beaumont, S.; Retailleau, P.; Dauban, P.; Dodd, R. H. Eur. J. Org. Chem. 2008, 5162.
- (176) Ilyn, A. P.; Loseva, M. V.; Tkachenko, S. E.; Vvedensky, V. Y.; Putsykina, E. B.; Kravchenko, D. V.; Khvat, A. V.; Krasavin, M. Y.;
- Ivachtchenko, A. V. J. Org. Chem. 2006, 71, 2811.
- (177) Zhang, J.; Jacobson, A.; Rusche, J. R.; Herlihy, W. J. Org. Chem. 1999, 64, 1074.
- (178) Mironov, M. A.; Ivantsova, M. N.; Mokrushin, V. S. Mol. Diversity 2003, 6, 193.
- (179) Rhoden, C. R. B.; Westermann, B.; Wessjohann, L. A. Synthesis 2008, 2077.
- (180) Tsirulnikov, S.; Nikulnikov, M.; Kysil, V.; Ivachtchenko, A.; Krasavin, M. Tetrahedron Lett. 2009, 50, 5529.
- (181) Balalaie, S.; Bararjanian, M.; Hosseinzadeh, S.; Rominger, F.; Bijanzadeh, H. R.; Wolf, E. Tetrahedron 2011, 67, 7294.

(182) (a) Huang, Y.; Dömling, A. Chem. Bio. Drug Design 2010, 76, 130. (b) Huang, Y.; Wolf, S.; Bista, M.; Meireles, L.; Camacho, C.;

Holak, T. A.; Dömling, A. Chem. Bio. Drug Design 2010, 76, 116.

(183) Nixey, T.; Tempest, P.; Hulme, C. Tetrahedron Lett. 2002, 43, 1637.

(184) Liu, H. X.; Dömling, A. J. Org. Chem. 2009, 74, 6895.

(185) Wang, W.; Herdtweck, E.; Dömling, A. Chem. Commun. 2010, 46, 770.

(186) Znabet, A.; Zonneveld, J.; Janssen, E.; De Kanter, F. J. J.; Helliwell, M.; Turner, N. J.; Ruijter, E.; Orru, R. V. A. Chem. Commun. 2010, 46, 7706.

(187) (a) Karpov, A. S.; Rominger, F.; Müller, T. J. J. Org. Biomol. Chem. 2005, 3, 4382. (b) Karpov, A. S.; Oeser, T.; Müller, T. J. J. Chem. Commun. 2004, 1502.

(188) Gupta, S.; Kumar, B.; Kundu, B. J. Org. Chem. 2011, 76, 10154. (189) (a) Ruijter, E.; Garcia-Hartjes, J.; Hoffmann, F.; van Wandelen,

L. T. M.; de Kanter, F. J. J.; Janssen, E.; Orru, R. V. A. Synlett 2010, 2485. (b) Bondzic, B. P.; Eilbracht, P. Org. Biomol. Chem. 2008, 6, 4059. (c) Airiau, E.; Spangenberg, T.; Girard, N.; Schoenfelder, A.; Salvadori, J.; Taddei, M.; Mann, A. Chem. Eur. J. 2008, 14, 10938.

(190) Wu, X.; Dai, X.; Nie, L.; Fang, H.; Chen, J.; Ren, Z.; Cao, W.; Zhao, G. Chem. Commun. 2010, 46, 2733.

(191) (a) Wu, X.; Dai, X.; Fang, H.; Nie, L.; Chen, J.; Cao, W.; Zhao, G. Chem.-Eur. J. 2011, 17, 10510. (b) Wu, X.; Fang, H.; Liu, Q.; Nie, L.; Chen, J.; Cao, W.; Zhao, G. Tetrahedron 2011, 67, 7251. (c) Rueping, M.; Volla, C. M. R.; Bolte, M.; Raabe, G. Adv. Synth. Catal. 2011, 353, 2853.

(192) Gonzalez-Vera, J. A.; Garcia-Lopez, M. T.; Herranz, R. J. Org. Chem. 2005, 70, 3660.

(193) Lu, K.; Luo, T.; Xiang, Z.; You, Z.; Fathi, R.; Chen, J.; Yang, Z. J. Comb. Chem. 2005, 7, 958.

(194) Ma, Z.; Xiang, Z.; Luo, T.; Lu, K.; Xu, Z.; Chen, J.; Yang, Z. J. Comb. Chem. 2006, 8, 696.

(195) El Kaim, L.; Grimaud, L.; Le Goff, X.-F.; Menes-Arzate, M.; Miranda, L. D. Chem. Commun. 2011, 47, 8145.

(196) Riguet, E. J. Org. Chem. 2011, 76, 8143.

(197) Rezaei, Z.; Firouzabadi, H.; Iranpoor, N.; Ghaderi, A.; Jafari, M.

R.; Jafari, A. A.; Zare, H. R. Eur. J. Med. Chem. 2009, 44, 4266. (198) Karageorge, G. N.; Macor, J. E. Tetrahedron Lett. 2011, 52, 5117.

(199) Cuny, G. D.; Hauske, J. R.; Hoemman, M. Z.; Chopra, I. U.S. Patent 2002/US6376670, 2002.

(200) (a) Gupta, S.; Sharma, S. K.; Mandadapu, A. K.; Gauniyal, H. M.; Kundu, B. Tetrahedron Lett. 2011, 52, 4288. (b) Liu, C.; Widenhoefer, R. A. J. Am. Chem. Soc. 2004, 126, 10250.

(201) Mironov, M. A.; Mokrushin, V. S.; Maltsev, S. S. Synlett 2003, 943.

(202) van Leusen, A. M.; Wildeman, J.; Oldenziel, O. H. J. Org. Chem. 1977, 42, 1153.

(203) (a) Sisko, J.; Kassick, A. J.; Mellinger, M.; Filan, J. J.; Allen, A.; Olsen, M. A. J. Org. Chem. 2000, 65, 1516. (b) Wang, L.; Woods, K. W.; Li, Q.; Barr, K. J.; McCroskey, R. W.; Hannick, S. M.; Gherke, L.;

Credo, R. B.; Hui, Y.-H.; Marsh, K.; Warner, R.; Lee, J. Y.; Zielinski-Mozng, N.; Frost, D.; Rosenberg, S. H.; Sham, H. L. J. Med. Chem.

2002, 45, 1697. (c) Beck, B.; Leppert, C. A.; Mueller, B. K.; Dömling, A. QSAR Comb. Sci. 2006, 25, 527.

(204) Porwal, S.; Kumar, R.; Maulik, P. R.; Chauhan, P. M. S. Tetrahedron Lett. 2006, 47, 5863.

(205) Kumar, S.; Kumar, V.; Chimni, S. S. Tetrahedron Lett. 2003, 44, 2101.

(206) Colombo, F.; Cravotto, G.; Palmisano, G.; Penoni, A.; Sisti, M. Eur. J. Org. Chem. 2008, 2801.

(207) Cravotto, G.; Giovenzana, G. B.; Maspero, A.; Pilati, T.; Penoni, A.; Palmisano, G. Tetrahedron Lett. 2006, 47, 6439.

(208) Abbiati, G.; Canevari, V.; Caimi, S.; Rossi, E. Tetrahedron Lett. 2005, 46, 7117.

(209) Shen, S. S.; Ji, S. J. Chin. J. Chem. 2008, 26, 935.

- (210) Zhu, H.; Stockigt, J.; Yu, Y.; Zou, H. Org. Lett. 2011, 13, 2792.
- (211) Majumder, S.; Bhuyan, P. J. Synlett 2011, 1547.

(212) Sharma, S. K.; Mandadapu, A. K.; Saifuddin, M.; Gupta, S.; Agarwal, P. K.; Mandwal, A. K.; Gauniyal, H. M.; Kundu, B. Tetrahedron Lett. 2010, 51, 6022.

(213) Huber, K.; Kast, O.; Bracher, F. Synthesis 2010, 3849.

(214) Turet, L.; Marko, I. E.; Tinant, B.; Declercq, J.-P.; Touillaux, R. Tetrahedron Lett. 2002, 43, 6591.

(215) Gubskaya, V. P.; Konovalova, N. P.; Nuretdinov, I. A.; Fazleeva, G. M.; Berezhnaya, L. S.; Sibgatullina, F. G.; Karaseva, I. P. Russ. Chem. Bull. 2002, 51, 1723.

(216) Lakshmi, N. V.; Thirumurugan, P.; Perumal, P. T. Tetrahedron Lett. 2010, 51, 1064.

(217) Babu, G.; Yu, H. M.; Yang, S. M.; Fang, J. M. Bioorg. Med. Chem. Lett. 2004, 14, 1129.

(218) Heda, L. C.; Sharma, R.; Pareek, C.; Chaudhari, P. B. E. J. Chem. 2009, 6, 770.

(219) Abelman, M. M.; Smith, S. C.; James, D. R. Tetrahedron Lett. 2003, 44, 4559.

(220) (a) Lavilla, R.; Gotsens, T.; Santano, M. C.; Bosch, J.; Camins, A.; Arnau, N.; Escubedo, E.; Camarasa, J.; Pallas, M. Bioorg. Chem. 1997, 25, 169. (b) Giorgi, G.; Adamo, M. F. A.; Ponticelli, F.; Ventura, A. Org. Biomol. Chem. 2010, 8, 5339. (c) Lebel, H.; Ladjel, C.; Bréthous, L. J. Am. Chem. Soc. 2007, 129, 13321.

(221) Manpadi, M.; Uglinskii, P. Y.; Rastogi, S. K.; Cotter, K. M.; Wong, Y. S. C.; Anderson, L. A.; Ortega, A. J.; Van Slambrouck, S.; Steelant, W. F. A.; Rogelj, S.; Tongwa, P.; Antipin, M. Y.; Magedov, I. V.; Kornienko, A. Org. Biomol. Chem. 2007, 5, 3865.

(222) Feng, L.; Xu, L.; Lam, K.; Zhou, Z.; Yip, C. W.; Chan, A. S. C. Tetrahedron Lett. 2005, 46, 8685.

(223) Kreher, R.; Wagner, P. H. Chem. Ber. 1980, 113, 3675.

(224) Slatt, J.; Romero, I.; Bergman, J. Synthesis 2004, 2760.

(225) (a) Zhao, K.; Xu, X. P.; Zhu, S. L.; Shi, D. Q.; Zhang, Y.; Ji, S. J. Synthesis 2009, 2697. (b) Thirumurugan, P.; Nandakumar, A.; Lakshmi, N. V.; Kumar, P. V. N. P.; Perumal, P. T. Magn. Reson. Chem. 2010, 48, 554.

(226) (a) Geng, L.-J.; Yu, J.-G.; Feng, G.-L. J. Chem. Res. (S) 2010, 333. (b) Geng, L.-J.; Feng, G.-L.; Yu, J.-G. J. Chem. Res. (S) 2010, 565. (c) Geng, L.-J.; Yu, J.-G.; Feng, G.-L.; Zhang, H.-L. J. Chem. Res. (S) 2011, 35, 74. (d) Geng, L.-J.; Yu, J.-G.; Feng, G.-L.; Zhang, H.-L.;

Zhang, Y.-M. Synth. Commun. 2011, 41, 3448. (e) Biradar, J. S.; Sharanbasappa, B. Green Chem. Lett. Rev. 2009, 2, 237.

(227) Chen, T.; Xu, X.-P.; Liu, H.-F.; Ji, S.-J. Tetrahedron 2011, 67, 5469.

(228) Thirumurugan, P.; Nandakumar, A.; Muralidharan, D.; Perumal, P. T. J. Comb. Chem. 2010, 12, 161.

(229) Thirumurugan, P.; Perumal, P. T. Tetrahedron Lett. 2009, 50, 4145.

(230) (a) Thirumurugan, P.; Perumal, P. T. Tetrahedron 2009, 65, 7620. (b) Thirumurugan, P.; Mahalaxmi, S.; Perumal, P. T. J. Chem. Sci. 2010, 122, 819.

(231) Sun, C.; Ji, S. J.; Liu, Y. Tetrahedron Lett. 2007, 48, 8987.

(232) Zhu, S. L.; Ji, S. J.; Su, X. M.; Sun, C.; Liu, Y. Tetrahedron Lett. 2008, 49, 1777.

(233) (a) Zhu, S. L.; Ji, S. J.; Zhao, K.; Liu, Y. Tetrahedron Lett. 2008, 49, 2578. (b) Quiroga, J.; Trilleras, J.; Sanchez, A. I.; Insuasty, B.; Abonia, R.; Nogueras, M.; Cobo, J. Lett. Org. Chem. 2009, 6, 381. (c) Quiroga, J.; Sanchez, A.; Cobo, J.; Glidewell, C. Acta Crystallogr., Sect. C: Cryst. Struct. Commun. 2009, 65, 374. (d) Low, J. N.; Cobo, J.; Sanchez, A.; Trilleras, J.; Glidewell, C. Acta Crystallogr., Sect. C: Cryst. Struct. Commun. 2007, 63, 287.

(234) Chen, T.; Xu, X. P.; Ji, S. J. J. Comb. Chem. 2010, 12, 659.

(235) (a) Lakshmi, N. V.; Thirumurugan, P.; Noorulla, K. M.; Perumal, P. T. Bioorg. Med. Chem. Lett. 2010, 20, 5054. (b) Nandakumar, A.; Thirumurugan, P.; Perumal, P. T.; Vembu, P.; Ponnuswamy, M. N.; Ramesh, P. Bioorg. Med. Chem. Lett. 2010, 20, 4252. (c) Zhao, K.; Zhu, S.-L.; Shi, D.-Q.; Xu, X.-P.; Ji, S.-J. Synthesis 2010, 1793.

(236) (a) Cheng, H.-G.; Chen, C.-B.; Tan, F.; Chang, N.-J.; Chen, J.-R.; Xiao, W.-J. Eur. J. Org. Chem. 2010, 4976. (b) Waser, J.; Gaspar, B.; Nambu, H.; Carreira, E. M. J. Am. Chem. Soc. 2006, 128, 11693.

Chemical Reviews Reviews Review Review

(237) Langer, P.; Doring, M.; Gorls, H. Eur. J. Org. Chem. 2001, 1511.

(238) (a) Damodiran, M.; Muralidharan, D.; Perumal, P. T. Bioorg. Med. Chem. Lett. **2009**, 19, 3611. (b) Barluenga, J.; Fañanás-Mastral, M.; Palomero, M. A.; Aznar, F.; Valdés, C. Chem. Eur. J. 2007, 13, 7682.

(239) (a) Touré, B. B.; Hall, D. G. Chem. Rev. 2009, 109, 4439. (b) Yamaguchi, R.; Hamasaki, T.; Sasaki, T.; Ohta, T.; Utimoto, K.; Kozima, S.; Takaya, H. J. Org. Chem. 1993, 58, 1136. (c) Yamaguchi, R.; Hamasaki, T.; Utimoto, K.; Kozima, S.; Takaya, H. Chem. Lett. 1990, 2161.

(240) (a) Tietze, L. F.; Rackelmann, N. Pure Appl. Chem. 2004, 76, 1967. (b) Tietze, L. F.; Modi, A. Med. Res. Rev. 2000, 20, 304. (c) Tietze, L. F.; Bachmann, J.; Schul, W. Angew. Chem., Int. Ed. 1988, 27, 971.

(241) Tietze, L. F.; Zhou, Y. F. Angew. Chem., Int. Ed. 1999, 38, 2045. (242) Deiters, A.; Pettersson, M.; Martin, S. F. J. Org. Chem. 2006, 71, 6547.

(243) (a) Mirabal-Gallardo, Y.; Soriano, M. D. P. C; Caballero, J.; Alzate-Morales, J.; Simirgiotis, M. J.; Santos, L. S. Synthesis 2012, 44, 144. (b) Ngouansavanh, T.; Zhu, J. Angew. Chem. Int. Ed. 2007, 46, 5775.

(244) Tietze, L. F.; Bachmann, J.; Wichmann, J.; Zhou, Y. F.; Raschke, T. Liebigs Ann./Recl. 1997, 881.

(245) Tietze, L. E.; Rackelmann, N.; Müller, I. Chem.-Eur. J. 2004, 10, 2722.

(246) Kong, Y. C.; Cheng, K. F.; Cambie, R. C.; Waterman, P. G. Chem. Commun. 1985, 47.

(247) (a) Ishikura, M.; Imaizumi, K.; Katagiri, N. Heterocycles 2000, 53, 2201. (b) Ishikura, M. Heterocycles 1995, 41, 1385. (c) Ishikura, M.; Imaizumi, K.; Katagiri, N. Heterocycles 2000, 53, 553.

(248) Xu, B.; Guo, Z. L.; Jin, W.-Y.; Wang, Z.-P.; Peng, Y.-G.; Guo, Q.-X. Angew. Chem., Int. Ed. 2012, 51, 1059.

(249) Beck, B.; Hess, S.; Dömling, A. Bioorg. Med. Chem. Lett. 2000, 10, 1701.

- (250) Choshi, T.; Yamada, S.; Sugino, E.; Kuwada, T.; Hibino, S. J. Org. Chem. 1995, 60, 5899.
- (251) Hirose, T.; Sunazuka, T.; Yamamoto, D.; Kaji, E.; Omura, S. Tetrahedron Lett. 2006, 47, 6761.

(252) Liu, J. F.; Ye, P.; Zhang, B.; Bi, G.; Sargent, K.; Yu, L.; Yohannes, D.; Baldino, C. M. J. Org. Chem. 2005, 70, 6339.

(253) Takiguchi, S.; Iizuka, T.; Kumakura, Y.-S.; Murasaki, K.; Ban, N.; Higuchi, K.; Kawasaki, T. J. Org. Chem. 2010, 75, 1126.

(254) (a) Kobayashi, T.; Nakashima, M.; Hakogi, T.; Tanaka, K.; Katsumura, S. Org. Lett. 2006, 8, 3809. (b) Kobayashi, T.; Takeuchi, K.; Miwa, J.; Tsuchikawa, H.; Katsumura, S. Chem. Commun. 2009, 3363.

(255) Sakaguchi, T.; Kobayashi, S.; Katsumura, S. Org. Biomol. Chem. 2011, 9, 257.

(256) Li, Y.; Kobayashi, T.; Katsumura, S. Tetrahedron Lett. 2009, 50, 4482.

(257) Ruff, B. M.; Zhong, S.; Nieger, M.; Sickert, M.; Schneider, C.; Bräse, S. Eur. J. Org. Chem. 2011, 6558.

(258) Jiang, B.; Yang, C.-G.; Gu, X.-H. Tetrahedron Lett. 2001, 42, 2545.

(259) Mitsudo, K.; Thansandote, P.; Wilhelm, T.; Mariampillai, B.; Lautens, M. Org. Lett. 2006, 8, 3939.

(260) Sunderhaus, J. D.; Dockendorff, C.; Martin, S. F. Org. Lett. 2007, 9, 4223.

(261) Ibad, M. F.; Hussain, M.; Abid, O. R.; Ali, A.; Ullah, I.; Zinad, D. S.; Langer, P. Synlett 2010, 411.

(262) Grigg, R.; Redpath, J.; Sridharan, V.; Wilson, D. Tetrahedron Lett. 1994, 35, 4429.

(263) Buscemi, G.; Miller, P. W.; Kealey, S.; Gee, A. D.; Long, N. J.; Passchier, J.; Vilar, R. Org. Biomol. Chem. 2011, 9, 3499.

(264) Ishikura, M.; Takahashi, N.; Yamada, K.; Yanada, R. Tetrahedron 2006, 62, 11580.

(265) (a) Chataigner, I.; Piettre, S. R. Org. Lett. 2007, 9, 4159. (b) Kuster, G. J. T.; Steeghs, R. H. J.; Scheeren, H. W. Eur. J. Org. Chem. 2001, 553.

(266) Blot, V.; Reissig, H. U. Synlett 2006, 2763.

(267) Beaumont, S.; Pons, V.; Retailleau, P.; Dodd, R. H.; Dauban, P. Angew. Chem., Int. Ed. 2010, 122, 1678.